

Section 6

Ethers

Diethyl ether

Alternative names

Ethyl ether, ethoxy ethane, ether, ethyl oxide, sulfuric ether, **not** petroleum ether

Reference codes

CAS number	60 29 7	Hazchem code	3YE
UN number	1155	EPA code	U117

Physical properties

Molecular weight	74	Cubic expansion coeff (per °C × 10 ³)	1.6
Empirical formula	C ₄ H ₁₀ O ₁	Surface tension (@20°C dyn/cm)	17
Boiling point (°C)	34.5	Absolute viscosity (@25°C cP)	0.24
Freezing point (°C)	-116	Refractive index (25°C)	1.352
Specific gravity (20/4)	0.715		

Fire hazards

Flash point (closed cup °C)	-45	Lower explosive limit (ppm)	18500
Autoignition temperature (°C)	160	Upper explosive limit (ppm)	360000
Electrical conductivity	3E-16		

Health hazards

IDLH (ppm)	19000	Vapour concentration @21°C ppm	610000
OES-TWA	400	Vapour density (relative to air)	2.57
OES-STEL	500	Vapour pressure @21°C mmHg	462
Odour threshold (ppm)	1	POCP	60

Aqueous effluent

Solubility in water (25°C %w/w)	6.9
Solubility of water in (25°C %w/w)	1.3
Log ₁₀ activated carbon partition	
Log ₁₀ partition in octanol/water (w/w)	+0.77
Biological oxygen demand w/w (days)	0.03 (5)
Theoretical oxygen demand w/w	2.59

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	6.98472
	B	1090.64
	C	231.20
Cox chart	A	7.00353
	B	1088.4

Solvent properties

Solubility parameter	7.4	Kauri butanol value	
Dipole (D)	1.3	Evaporation time (ether = 1)	1.0
Dielectric constant (20°C)	4.3	Evaporation time (BuAc = 1)	28.0
Polarity (water 100)	11.7		

Thermal information

Latent heat (cal/mol)	6216
Nett heat of combustion (kcal/gmol)	598
Specific heat (cal/mol/°C)	40
Critical pressure (MN/m ²)	3.61
Critical temperature (K)	473
Latent heat of fusion (cal/mol)	1735
Van der Waals' volume	3.39
Van der Waals' surface area	3.02
Molar volume	103.5

Solute	Azeotrope		Reference	Solute γ^m	Reference	Partition coefficient	Reference
	X% w/w	°C					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane	68	33	8296				
<i>n</i> -Hexane	None		8301				
<i>n</i> -Heptane							
<i>n</i> -Octane				1.5	1x/3/1077		
<i>n</i> -Nonane							
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane							
Benzene	None		8293	0.9	3+4/516		
Toluene	None		8304	1.2	1x/3/1077		
Ethylbenzene							
Xylenes							
C ₉ Aromatics							
Tetralin							
<i>Alcohols</i>							
Methanol	2	62		4.8	2a/170	2.9	P157
Ethanol	None		4029	3.8	2a/375	1.75	V2/341
<i>n</i> -Propanol	None		6464			0.18	P678
<i>i</i> -Propanol	None		6351			0.44	P649
<i>n</i> -Butanol	None		8104				
<i>i</i> -Butanol							
<i>s</i> -Butanol							
<i>n</i> -Amyl alc.							
<i>i</i> -Amyl alc.						1.3	P1274
Cyclohexanol							
1-Octanol							
Ethanediol							
DEG						55.7	P1004
1,2-Propanediol						13.4	P657
<i>Glycol ethers</i>							
PGME							
EGME						1.61	P655
EEE						1.22	P1001
EGBE							
<i>Chlorinated</i>							
MDC	30	41	1566	0.6	3+4/492		
Chloroform	None		1474	0.4	3+4/486		
Carbon tet.	None		1136				
1,2-EDC	None		2987				
1,1,1-TCA							
TCE							
Perk.							
MCB							

Ethers

Solute	Azeotrope		Solute γ°	Reference	Partition coefficient	Reference
	X% w/w	$^{\circ}\text{C}$				
<i>Ketones</i>						
Acetone	None		2.2	3+4/177	0.28	V2/470
MEK			1.9	1x/3/1077		
MIBK						
Cyclohexanone						
NMP						
Acetophenone					<0.01	P2744
<i>Ethers</i>						
Diethyl ether	-		-	-	-	-
DIPE						
Dibutyl ether	None					8305
MTBE						
1,4-Dioxane			2.0	1x/3/1077		
THF						
<i>Esters</i>						
Me acetate	None				0.08	P516
Et acetate			1.1	3+4/513	0.03	P860
<i>i</i> -Propyl acetate						
<i>n</i> -Butyl acetate						
Cellosolve acetate						
<i>Miscellaneous</i>						
DMF					24.3	P594
DMAc						
DMSO						
Sulfolane						
CS ₂	99	34	1.6	3+4/495		
Acetic acid	None		2.6	3+4/502	0.41	V2/228
Aniline					0.03	P1710
Nitrobenzene	None					8297
Morpholine						
Pyridine					0.20	P1101
2-Nitropropane						
Acetonitrile	None		2.5	3+4/499	0.40	P261
Furfuraldehyde						
Phenol					<0.01	P1617
Water	99	34	28.6	1a/257		

Diisopropyl ether

Alternative names

Isopropyl ether, DIPE

Reference codes

CAS number	108 20 3	Hazchem code	3YE
UN number	1159	EPA code	

Physical properties

Molecular weight	102	Cubic expansion coeff (per °C × 10 ³)	1.4
Empirical formula	C ₆ H ₁₄ O ₁	Surface tension (@20°C dyn/cm)	18
Boiling point (°C)	68	Absolute viscosity (@25°C cP)	0.33
Freezing point (°C)	-86	Refractive index (25°C)	1.367
Specific gravity (20/4)	0.724		

Fire hazards

Flash point (closed cup °C)	-28	Lower explosive limit (ppm)	14000
Autoignition temperature (°C)	430	Upper explosive limit (ppm)	79000
Electrical conductivity			

Health hazards

IDLH (ppm)	10000	Vapour concentration@ 21°C ppm	193000
OES-TWA	250	Vapour density (relative to air)	3.58
OES-STEL	310	Vapour pressure @21°C mmHg	123
Odour threshold (ppm)	0.1	POCP	

Aqueous effluent

Solubility in water (25°C %w/w)	1.2
Solubility of water in (25°C %w/w)	0.62
Log ₁₀ activated carbon partition	2.9
Log ₁₀ partition in octanol/water (w/w)	+2.0
Biological oxygen demand w/w (days)	0.19 (5)
Theoretical oxygen demand w/w	2.83

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	6.84953
	B	1139.34
	C	231.742
Cox chart	A	7.09624
	B	1256.2

Solvent properties

Solubility parameter	6.9	Kauri butanol value	
Dipole (D)	1.2	Evaporation time (ether = 1)	1.6
Dielectric constant (20°C)		Evaporation time (BuAc = 1)	8.1
Polarity (water 100)			

Thermal information

Latent heat (cal/mol)	6936
Nett heat of combustion (kcal/gmol)	885
Specific heat (cal/mol/°C)	52
Critical pressure (MN/m ²)	3.14
Critical temperature (K)	500
Latent heat of fusion (cal/mol)	2631
Van der Waals' volume	4.74
Van der Waals' surface area	4.09
Molar volume	142.3

Solute	Azeotrope		Reference	Solute γ^*	Reference	Partition coefficient	Reference
	X% w/w	°C					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane							
<i>n</i> -Hexane	53	67	12128				
<i>n</i> -Heptane	None		12196	1.1	3+4/559		
<i>n</i> -Octane				1.3	1x/3/1243		
<i>n</i> -Nonane							
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane				1.0	3+4/555		
Benzene	None		10863	1.2	3+4/553		
Toluene				1.1	3+4/558		
Ethylbenzene				1.4	3+4/563		
Xylenes							
C ₉ Aromatics							
Tetralin							
<i>Alcohols</i>							
Methanol	76	57	2091a	3.3	2a/261		
Ethanol	83	64	4110	5.0	2a/459		
<i>n</i> -Propanol				3.5	2a/586		
<i>i</i> -Propanol	85	66	6391	4.5	2b/101	0.5	V2/618
<i>n</i> -Butanol				3.0	2b/202		
<i>i</i> -Butanol							
<i>s</i> -Butanol							
<i>n</i> -Amyl alc.							
<i>i</i> -Amyl alc.							
Cyclohexanol							
1-Octanol							
Ethenediol							
DEG							
1,2-Propanediol							
<i>Glycol ethers</i>							
PGME							
EGME							
EEE							
EGBE							
<i>Chlorinated</i>							
MDC							
Chloroform	64	71	1496	0.5	3+4/537		
Carbon tet.	None		1163	1.1	3+4/529		
1,2-EDC	None		3004				
1,1,1-TCA							
TCE	None		2331				
Perk.							
MCB							

Ethers

Solute	Azeotrope		Reference	Solute γ^{∞}	Reference	Partition coefficient	Reference
	X% w/w	°C					
<i>Ketones</i>							
Acetone	39	54	5386				
MEK	Azeo		3b/357	2.8	3b/357		
MIBK							
Cyclohexanone				1.7	3b/506		
NMP							
Acetophenone							
<i>Ethers</i>							
Diethyl ether							
DIPE	-		-	-	-	-	-
Dibutyl ether							
MTBE							
1,4-Dioxane	None		7548	2.2	1x/3/1243		
THF							
<i>Esters</i>							
Me acetate							
Et acetate							
<i>i</i> -Propyl acetate							
<i>n</i> -Butyl acetate							
Cellosolve acetate							
<i>Miscellaneous</i>							
DMF							
DMAc						4.5	V3/103
DMSO							
Sulfolane							
CS ₂							
Acetic acid	None		3185	3.2	3+4/544	0.6	V2/282
Aniline							
Nitrobenzene							
Morpholine							
Pyridine							
2-Nitropropane							
Acetonitrile							
Furfuraldehyde							
Phenol						0.01	P1656
Water	95	62	579	19.2	1/525		

Dibutyl ether

Alternative names

Butyl ether

Reference codes

CAS number 142 96 1 Hazchem code
UN number EPA code

Physical properties

Molecular weight	130	Cubic expansion coeff (per °C × 10 ³)	1.15
Empirical formula	C ₈ H ₁₈ O ₁	Surface tension (@20°C dyn/cm)	1.4
Boiling point (°C)	142	Absolute viscosity (@25°C cP)	0.63
Freezing point (°C)	-95	Refractive index (25°C)	1.397
Specific gravity (20/4)	0.769		

Fire hazards

Flash point (closed cup °C)	25	Lower explosive limit (ppm)	15000
Autoignition temperature (°C)	194	Upper explosive limit (ppm)	76000
Electrical conductivity			

Health hazards

IDLH (ppm)		Vapour concentration @21°C ppm	7377
OES-TWA		Vapour density (relative to air)	4.48
OES-STEL		Vapour pressure @21°C mmHg	5.5
Odour threshold (ppm)	0.5	POCP	

Aqueous effluent

Solubility in water (25°C %w/w)	0.03
Solubility of water in (25°C %w/w)	0.20
Log ₁₀ activated carbon partition	4.59
Log ₁₀ partition in octanol/water (w/w)	
Biological oxygen demand w/w (days)	
Theoretical oxygen demand w/w	2.95

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	6.7963
	B	1297.29
	C	191.03
Cox chart	A	7.31357
	B	1649.0

Solvent properties

Solubility parameter	7.2	Kauri butanol value	
Dipole (D)	1.2	Evaporation time (ether = 1)	
Dielectric constant (20°C)		Evaporation time (BuAc = 1)	
Polarity (water 100)	7.1		

Thermal information

Latent heat (cal/mol)	8944
Nett heat of combustion (kcal/gmol)	1182
Specific heat (cal/mol/°C)	66.0
Critical pressure (MN/m ²)	
Critical temperature (K)	307
Latent heat of fusion (cal/mol)	
Van der Waals' volume	6.09
Van der Waals' surface area	5.18
Molar volume	170.4

Solute	Azeotrope		Reference	Solute γ^*	Reference	Partition coefficient	Reference
	X% w/w	°C					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane				1.0	1x/3/1370		
<i>n</i> -Hexane				1.0	1x/3/1370		
<i>n</i> -Heptane				1.0	1x/3/1370		
<i>n</i> -Octane				1.1	1x/3/1370		
<i>n</i> -Nonane							
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane				1.0	1x/3/1370		
Benzene				0.9	1x/3/1370		
Toluene				1.0	1x/3/1370		
Ethylbenzene	None		14102				
Xylenes	78	142	14117				
C ₉ Aromatics							
Tetralin							
<i>Alcohols</i>							
Methanol							
Ethanol	None		4170	3.5	2e/391	0.98	V2/389
<i>n</i> -Propanol							
<i>i</i> -Propanol							
<i>n</i> -Butanol	17	118	8195	0.7	2d/231	0.04	V3/122
<i>i</i> -Butanol	None		8379				
<i>s</i> -Butanol							
<i>n</i> -Amyl alc.	50	135	9770				
<i>i</i> -Amyl alc.	35	130	9883				
Cyclohexanol	None		11745				
1-Octanol							
Ethenediol	90	140	4354				
DEG							
1,2-Propanediol		136	6668				
<i>Glycol ethers</i>							
PGME	63	138	10003				
EGME	32	122	6615				
EEE	50	127	8479				
EGBE	None		12247				
<i>Chlorinated</i>							
MDC							
Chloroform	None		1501a	0.5	1x/3/1370		
Carbon tet.				0.8	1x/3/1370		
1,2-EDC				1.2	1x/3/1370		
1,1,1-TCA				0.8	1x/3/1370		
TCE							
Perk.							
MCB	None		10537				

Solute	Azeotrope		Solute γ^{∞}	Reference	Partition coefficient	Reference
	X% w/w	°C				
<i>Ketones</i>						
Acetone			2.3	1x/3/1370	0.07	CEP
MEK						
MIBK						
Cyclohexanone						
NMP						
Acetophenone						
<i>Ethers</i>						
Diethyl ether	None					8305
DIPE						
Dibutyl ether	-		-	-	-	-
MTBE						
1,4-Dioxane			2.1	1x/3/1370		
THF						
<i>Esters</i>						
Me acetate						
Et acetate						
<i>i</i> -Propyl acetate						
<i>n</i> -Butyl acetate	5	126				11333
Cellosolve acetate						
<i>Miscellaneous</i>						
DMF						
DMAc					5.0	V3/104
DMSO						
Sulfolane						
CS ₂						
Acetic acid	None				1.0	V2/304
Aniline	None					3220 11198
Nitrobenzene						
Morpholine						
Pyridine						
2-Nitropropane						
Acetonitrile						
Furfuraldehyde	80	138	4.2	3a/139		8788
Phenol	None					10960
Water	67	93				735

Methyl tert butyl ether

Alternative names

Tert butyl methyl ether, MTBE

Reference codes

CAS number 1634 04 4 Hazchem code
UN number EPA code

Physical properties

Molecular weight	88	Cubic expansion coeff (per °C × 10 ³)	
Empirical formula	C ₅ H ₁₂ O ₁	Surface tension (@20°C dyn/cm)	18.3
Boiling point (°C)	55	Absolute viscosity (@25°C cP)	0.35
Freezing point (°C)	-109	Refractive index (20°C)	1.369
Specific gravity (20/4)	0.741		

Fire hazards

Flash point (closed cup °C)	-34	Lower explosive limit (ppm)	16000
Autoignition temperature (°C)	460	Upper explosive limit (ppm)	84000
Electrical conductivity			

Health hazards

IDLH (ppm)		Vapour concentration @21°C ppm	372000
OES-TWA		Vapour density (relative to air)	3.06
OES-STEL		Vapour pressure @21°C mmHg	206
Odour threshold (ppm)		POCP	

Aqueous effluent

Solubility in water (25°C %w/w)	4.3
Solubility of water in (25°C %w/w)	1.4
Log ₁₀ activated carbon partition	
Log ₁₀ partition in octanol/water (w/w)	
Biological oxygen demand w/w (days)	
Theoretical oxygen demand w/w	2.75

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	
	B	
	C	
Cox chart	A	7.06046
	B	1191.2

Solvent properties

Solubility parameter	7.4	Kauri butanol value	
Dipole (D)	1.2	Evaporation time (ether = 1)	1.6
Dielectric constant (20°C)	4.5	Evaporation time (BuAc = 1)	8.4
Polarity (water 100)	14.8		

Thermal information

Latent heat (cal/mol)	7030
Nett heat of combustion (kcal/gmol)	740
Specific heat (cal/mol/°C)	
Critical pressure (MN/m ²)	
Critical temperature (K)	
Latent heat of fusion (cal/mol)	
Van der Waals' volume	4.07
Van der Waals' surface area	3.63
Molar volume	119.0

Solute	Azeotrope		Reference	Solute γ°	Reference	Partition coefficient	Reference
	X% w/w	$^{\circ}\text{C}$					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane				1.2	1x/3/1152		
<i>n</i> -Hexane				0.9	1x/3/1152		
<i>n</i> -Heptane							
<i>n</i> -Octane							
<i>n</i> -Nonane							
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane				1.1	1x/3/1152		
Benzene							
Toluene							
Ethylbenzene							
Xylenes							
C ₉ Aromatics							
Tetralin							
<i>Alcohols</i>							
Methanol	90	51	2058	3.0	2c/160		
Ethanol				3.0			
<i>n</i> -Propanol				2.9			
<i>i</i> -Propanol				2.6			
<i>n</i> -Butanol							
<i>i</i> -Butanol							
<i>s</i> -Butanol							
<i>n</i> -Amyl alc.							
<i>i</i> -Amyl alc.							
Cyclohexanol							
1-Octanol							
Ethenediol							
DEG							
1,2-Propanediol							
<i>Glycol ethers</i>							
PGME							
EGME							
EEE							
EGBE							
<i>Chlorinated</i>							
MDC							
Chloroform							
Carbon tet.							
1,2-EDC							
1,1,1-TCA							
TCE							
Perk.							
MCB							

Ethers

Solute	Azeotrope		Solute γ^{∞}	Reference	Partition coefficient	Reference
	X% w/w	°C				
<i>Ketones</i>						
Acetone						
MEK						
MIBK						
Cyclohexanone						
NMP						
Acetophenone						
<i>Ethers</i>						
Diethyl ether						
DIPE						
Dibutyl ether						
MTBE	-		-		-	-
1,4-Dioxane						
THF						
<i>Esters</i>						
Me acetate						
Et acetate						
<i>i</i> -Propyl acetate						
<i>n</i> -Butyl acetate						
Cellosolve acetate						
<i>Miscellaneous</i>						
DMF						
DMAc						
DMSO						
Sulfolane						
CS ₂						
Acetic acid						
Aniline						
Nitrobenzene						
Morpholine						
Pyridine						
2-Nitropropane						
Acetonitrile						
Furfuraldehyde						
Phenol						
Water	97	52	466			

1,4-Dioxane

Alternative names

Glycol ethylene ether, p-dioxane, diethylene dioxide, diethylene oxide

Reference codes

CAS number	123 91 1	Hazchem code	2SE
UN number	1185	EPA code	

Physical properties

Molecular weight	88	Cubic expansion coeff (per °C × 10 ³)	1.1
Empirical formula	C ₄ H ₈ O ₂	Surface tension (@20°C dyn/cm)	40
Boiling point (°C)	101	Absolute viscosity (@25°C cP)	1.3
Freezing point (°C)	+12	Refractive index (25°C)	1.420
Specific gravity (20/4)	1.034		

Fire hazards

Flash point (closed cup °C)	12	Lower explosive limit (ppm)	20000
Autoignition temperature (°C)	180	Upper explosive limit (ppm)	222000
Electrical conductivity	5E-15		

Health hazards

IDLH (ppm)	200	Vapour concentration @21°C ppm	41000
OES-TWA	25	Vapour density (relative to air)	3.06
OES-STEL	100	Vapour pressure @21°C mmHg	32
Odour threshold (ppm)	170	POCP	

Aqueous effluent

Solubility in water (25°C %w/w)	Total
Solubility of water in (25°C %w/w)	Total
Log ₁₀ activated carbon partition	
Log ₁₀ partition in octanol/water (w/w)	-0.42
Biological oxygen demand w/w (days)	0
Theoretical oxygen demand w/w	1.82

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	7.43155
	B	1554.679
	C	240.337
Cox chart	A	7.19047
	B	1426.5

Solvent properties

Solubility parameter	10.0	Kauri butanol value	
Dipole (D)	0.4	Evaporation time (ether = 1)	7.3
Dielectric constant (20°C)	2.21	Evaporation time (BuAc = 1)	2.2
Polarity (water 100)	16.4		

Thermal information

Latent heat (cal/mol)	8510
Nett heat of combustion (kcal/gmol)	567
Specific heat (cal/mol/°C)	36
Critical pressure (MN/m ²)	5.21
Critical temperature (K)	588
Latent heat of fusion (cal/mol)	3080
Van der Waals' volume	3.19
Van der Waals' surface area	2.64
Molar volume	85.1

Solute	Azeotrope		Reference	Solute γ^*	Reference	Partition coefficient	Reference
	X% w/w	$^{\circ}\text{C}$					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane							
<i>n</i> -Hexane	2	60	7547	3.3	3+4/472		
<i>n</i> -Heptane	44	92	7552	4.0	1x/3/1052		
<i>n</i> -Octane		100	7554	7.4	1x/3/1052		
<i>n</i> -Nonane				4.7	3+4/481		
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane	25	80	7540	2.6	3+4/468		
Benzene	12	82	7537	1.1	3+4/465		
Toluene	80	102	7550	1.2	1x/3/1052		
Ethylbenzene							
Xylenes							
C ₉ Aromatics							
Tetralin							
<i>Alcohols</i>							
Methanol	None		1998	1.6	2a/148		
Ethanol	9	78	4011	2.5	1x/3/1052		
<i>n</i> -Propanol	45	95	6447	1.7	2a/531		
<i>i</i> -Propanol	None		6337	1.2	2b/56		
<i>n</i> -Butanol	None		7519	1.5	2b/147		
<i>i</i> -Butanol	96	101	7522	1.4	2b/278		
<i>s</i> -Butanol	60	99	7520				
<i>n</i> -Amyl alc.							
<i>i</i> -Amyl alc.							
Cyclohexanol	None		7543				
1-Octanol							
Ethanediol	None		4206				
DEG							
1,2-Propanediol							
<i>Glycol ethers</i>							
PGME							
EGME	None		6541				
EEE							
EGBE							
<i>Chlorinated</i>							
MDC							
Chloroform	None		1465	0.46	3+4/441		
Carbon tet.	None		1124	1.2	1x/3/1052		
1,2-EDC	None		2979	1.4	3+4/447		
1,1,1-TCA							
TCE	None		2301				
Perk.	None		2181				
MCB							

Ethers

Solute	Azeotrope		Solute γ^{∞}	Reference	Partition coefficient	Reference
	X% w/w	°C				
<i>Ketones</i>						
Acetone	None			5333		
MEK			1.3		1x/3/1052	
MIBK			1.6		3b/523	
Cyclohexanone	None			7539		
NMP						
Acetophenone						
<i>Ethers</i>						
Diethyl ether						
DIPE	None			7548		
Dibutyl ether						
MTBE						
1,4-Dioxane	-		-	-	-	-
THF						
<i>Esters</i>						
Me acetate						
Et acetate	None		1.1	7514	3+4/455	
<i>i</i> -Propyl acetate	None			7527		
<i>n</i> -Butyl acetate						
Cellosolve acetate						
<i>Miscellaneous</i>						
DMF			1.7		3+4/454	
DMAc						
DMSO			2.2		3+4/450	
Sulfolane						
CS ₂			2.4		3+4/446	
Acetic acid	23	119	2.3	119	3+4/448	
Aniline						
Nitrobenzene						
Morpholine						
Pyridine	None			7524		
2-Nitropropane						
Acetonitrile	None			2769a		
Furfuraldehyde						
Phenol						
Water	82	88	4.1	349	1/382	

Tetrahydrofuran

Alternative names

THF, tetramethylene oxide, 1,4-epoxybutane, oxacyclopentane

Reference codes

CAS number	109 99 9	Hazchem code	2SE
UN number	2056	EPA code	U213

Physical properties

Molecular weight	72	Cubic expansion coeff (per °C × 10 ³)	1.1
Empirical formula	C ₄ H ₈ O ₁	Surface tension (@20°C dyn/cm)	28
Boiling point (°C)	66	Absolute viscosity (@25°C cP)	0.55
Freezing point (°C)	-109	Refractive index (25°C)	1.404
Specific gravity (20/4)	0.888		

Fire hazards

Flash point (closed cup °C)	-15	Lower explosive limit (ppm)	23000
Autoignition temperature (°C)	212	Upper explosive limit (ppm)	118000
Electrical conductivity	4.5 E-5		

Health hazards

IDLH (ppm)		Vapour concentration @21°C ppm	230000
OES-TWA	100	Vapour density (relative to air)	2.5
OES-STEL	200	Vapour pressure @21°C mmHg	133
Odour threshold (ppm)	30	POCP	70

Aqueous effluent

Solubility in water (25°C %w/w)		Total	Lower critical solution
Solubility of water in (25°C %w/w)		Total	temperature 72°C
Log ₁₀ activated carbon partition			
Log ₁₀ partition in octanol/water (w/w)		+0.46	
Biological oxygen demand w/w (days)			
Theoretical oxygen demand w/w		2.59	

Vapour pressure equation constants (Log₁₀, mmHg)

Antoine equation	A	6.99515
	B	1202.29
	C	226.254
Cox chart	A	7.09092
	B	1246.2

Solvent properties

Solubility parameter	9.1	Kauri butanol value	
Dipole (D)	1.75	Evaporation time (ether = 1)	2.3
Dielectric constant (20°C)	7.6	Evaporation time (BuAc = 1)	6.3
Polarity (water 100)	21		

Thermal information

Latent heat (cal/mol)	6664
Nett heat of combustion (kcal/gmol)	601
Specific heat (cal/mol/°C)	36
Critical pressure (MN/m ²)	5.2
Critical temperature (K)	541
Latent heat of fusion (cal/mol)	
Van der Waals' volume	2.94
Van der Waals' surface area	2.72
Molar volume	81.08

Solute	Azeotrope		Reference	Solute γ^*	Reference	Partition coefficient	Reference
	X% w/w	°C					
<i>Hydrocarbons</i>							
<i>n</i> -Pentane	None			2.0	1x/3/1046		
<i>n</i> -Hexane	50	63	7407	1.9	1x/3/1046		
<i>n</i> -Heptane							
<i>n</i> -Octane				1.7	1x/3/1047		
<i>n</i> -Nonane							
<i>n</i> -Decane							
2,2,4-TMP							
Cyclohexane	97	60		1.7	1x/3/1046		
Benzene				0.84	1x/3/1046		
Toluene				0.85	1x/3/1046		
Ethylbenzene				0.90	1x/3/1047		
Xylenes							
C ₉ Aromatics				0.9	1x/3/1047		
Tetralin							
<i>Alcohols</i>							
Methanol	69	61	1996	2.4	2a/141		
Ethanol	90	66	4009	1.9	2c/328		
<i>n</i> -Propanol	None			1.4	2c/497		
<i>i</i> -Propanol	None		6335a	1.4	2b/55		
<i>n</i> -Butanol	None			1.2	2b/146		
<i>i</i> -Butanol							
<i>s</i> -Butanol							
<i>n</i> -Amyl alc.							
<i>i</i> -Amyl alc.							
Cyclohexanol							
1-Octanol							
Ethanediol	None		4204a	6.6	2d/3		
DEG							
1,2-Propanediol							
<i>Glycol ethers</i>							
PGME							
EGME							
EEE							
EGBE							
<i>Chlorinated</i>							
MDC				0.5	1x/1/100		
Chloroform	34	72	1464	0.25	1x/3/1046		
Carbon tet.	None			0.75	3+4/429		
1,2-EDC				0.60	1x/3/1046		
1,1,1-TCA							
TCE							
Perk.							
MCB							

Ethers

Solute	Azeotrope		Solute γ^{∞}	Reference	Partition coefficient	Reference
	X% w/w	°C				
<i>Ketones</i>						
Acetone	8	64				
MEK						
MIBK						
Cyclohexanone						
NMP						
Acetophenone						
<i>Ethers</i>						
Diethyl ether						
DIPE						
Dibutyl ether						
MTBE						
1,4-Dioxane			1.1	1x/3/1046		
THF	-		-	-	-	-
<i>Esters</i>						
Me acetate						
Et acetate			1.1	1x/3/1046		
<i>i</i> -Propyl acetate						
<i>n</i> -Butyl acetate						
Cellosolve acetate						
<i>Miscellaneous</i>						
DMF						
DMAc						
DMSO	None		4.6	3+4/433	2.05	V2/396
Sulfolane						
CS ₂						
Acetic acid						
Aniline						
Nitrobenzene						
Morpholine						
Pyridine						
2-Nitropropane	None					6273
Acetonitrile						
Furfuraldehyde						
Phenol						
Water	96	64		10.4	1/367	345