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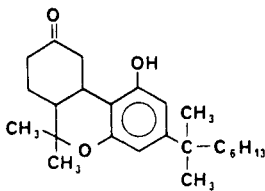
NABILONE

Therapeutic Function: Antianxiety

Chemical Name: 1-Hydroxy-3-(1',1'-dimethylheptyl)-6,6-dimethyl-6,6a,7,8,10,10a-hexahydro-9H-dibenzo[b,d]pyran-9-one

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 51022-71-0

Trade Name	Manufacturer	Country	Year Introduced
Cesamet	Lilly	Canada	1982
Cesametic	Lilly	W. Germany	1983
Cesamet	Lilly	U.K.	1983

Raw Materials

dl-3-(1',1'-Dimethylheptyl)-6,6a,7,8-tetrahydro-1-hydroxy-6,6-dimethyl-9H-dibenzo[b,d]pyran-9-one

Lithium
Ammonia

Manufacturing Process

A solution of 1.5 g of dl-3-(1',1'-dimethylheptyl)-6,6a,7,8-tetrahydro-1-hydroxy-6,6-dimethyl-9H-dibenzo[b,d]pyran-9-one in 50 ml of anhydrous tetrahydrofuran (THF) was added dropwise to a solution of lithium metal in liquid ammonia at -80°C . Excess lithium metal was added in chunks to the solution as the blue color, indicating free dissolved lithium, disappeared. After the addition was complete, ammonium chloride was added to react with any excess lithium metal still present.

The mixture was then allowed to warm to room temperature in a nitrogen atmosphere during which process the ammonia evaporated. The reaction mixture was then acidified with 1 N aqueous hydrochloric acid, and the organic constituents extracted with ethyl acetate. The ethyl acetate extracts were combined, washed with water and dried. Evaporation of the ethyl acetate under reduced pressure yielded 1.4 g of crude dl-trans-3-(1',1'-dimethylheptyl)-6,6a β ,7,8,10,10a β -hexahydro-1-hydroxy-6,6-dimethyl-9H-dibenzo[b,d]pyran-9-one. The

crude product was chromatographed over 50 g of silica gel from benzene solution and the desired product was eluted in 20 ml fractions with a benzene eluant containing 2% ethyl acetate. Fractions 200 to 240 contained 808 mg of a white crystalline solid comprising purified dl-trans-3-(1',1'-dimethylheptyl)-6,6a β ,7,8,10,10a β -hexahydro-1-hydroxy-6,6-dimethyl-9H-dibenzo[b,d]pyran-9-one. The purified compound melted at 159°C to 160°C after recrystallization from an ethyl acetate-hexane solvent mixture.

References

Merck Index 6193

DFU 3 (3) 207 (1978)

OCDS Vol. 3, p 189 (1984)

DOT 19 (7) 415 & (8) 436 (1983)

I.N. p. 652

Archer, R.A.; U.S. Patents 3,928,598; December 23, 1975; 3,944,673; March 16, 1976; and 3,953,603; April 27, 1976; all assigned to Eli Lilly & Co.

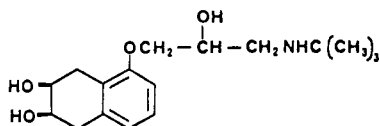
NADOLOL

Therapeutic Function: Antiarrhythmic

Chemical Name: 2,3-Cis-1,2,3,4-tetrahydro-5-[2-hydroxy-3-(tert-butylamino)propoxy]-2,3-naphthalenediol

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 42200-33-9

Trade Name	Manufacturer	Country	Year Introduced
Solgol	Heyden	W. Germany	1978
Corgard	Squibb	Switz.	1978
Corgard	Squibb	U.K.	1979
Corgard	Squibb	U.S.	1979
Corgard	Squibb	Italy	1980
Corgard	Squibb	France	1982
Betadol	Fako	Turkey	—
Corzide	Squibb	U.S.	—

Raw Materials

5,8-Dihydro-1-naphthol	Acetic anhydride
Silver acetate	Iodine
Sodium hydroxide	Sodium methoxide
Epichlorohydrin	tert-Butylamine

Manufacturing Process

(a) *cis*-5,6,7,8-Tetrahydro-1,6,7-naphthalenetriol: A solution of 29.2 g (0.2 mol) of 5,8-dihydro-1-naphthol and 40 ml of acetic anhydride in 100 ml of pyridine is prepared. After 16

hours the solvent is removed in vacuo and the residue dissolved in ether and washed with 200 ml of 5% hydrochloric acid, water, 200 ml of 10% sodium hydroxide, saturated salt solution and dried. Solvent removal gives 34.2 g (90.5%) of crude acetate which is dissolved in 900 ml of acetic acid and 36 ml of water. 53.3 g (0.32 mol) of silver acetate is added followed by 40.6 g (0.16 g-atom) of iodine. The slurry is heated with good stirring at $85^{\circ}\pm 10^{\circ}\text{C}$ for 3 hours under nitrogen, cooled and filtered. The filtrate is evaporated in vacuo and the residue dissolved in 250 ml of methanol and cooled to 0°C .

A solution of 40 g of sodium hydroxide in 200 ml of water is added under nitrogen and the mixture stirred overnight. The bulk of the methanol is removed in vacuo whereupon a solid forms. The solid is separated by filtration, dissolved in 150 ml of water and acidified with 20 ml of concentrated hydrochloric acid. Cooling gives a solid which is filtered and dried to give 16.5 g *cis*-5,6,7,8-tetrahydro-1,6,7-naphthalenetriol, melting point 184.5°C to 187°C . Three recrystallizations from absolute ethanol give the analytical sample, melting point 188°C to 188.5°C .

(b) 2,3-cis-1,2,3,4-Tetrahydro-5-[2,3-(epoxy)propoxy]-2,3-naphthalenediol: A solution of 1.20 g (0.03 mol) of sodium methoxide and 5.4 g (0.03 mol) of *cis*-5,6,7,8-tetrahydro-1,6,7-naphthalenetriol in 200 ml of methanol is prepared under nitrogen. The residue obtained upon solvent removal is stirred overnight with 200 ml of dimethylsulfoxide and 4.65 g (0.05 mol) of epichlorohydrin under nitrogen. The bulk of the solvent is removed at 50°C at 0.1 mm and the residue dissolved in 100 ml of water. Extraction with chloroform (10 x 200 ml) gives a solid which is recrystallized from 150 ml of hexane-ethyl acetate to give epoxy diol of the above title.

(c) 2,3-cis-1,2,3,4-Tetrahydro-5-[2-hydroxy-3-(tert-butylamino)propoxy]-2,3-naphthalenediol: A mixture of 2,3-*cis*-1,2,3,4-tetrahydro-5-[2,3-(epoxy)propoxy]-2,3-naphthalenediol (melting point 104°C to 107°C , one spot on TLC-alumina, 5% methanol in chloroform, iodine visualization) and 22 ml of *tert*-butylamine is heated at 85°C to 95°C for 15 hours in a Parr bomb and the excess amine removed in vacuo. The solid obtained by trituration of the residue with ether is filtered and recrystallized from benzene to give 3.4 g, melting point 124°C to 136°C .

References

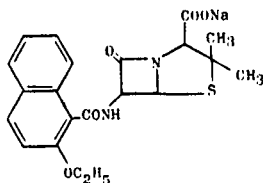
- Merck Index 6195
 DFU 1 (9) 434 (1976)
 Kleeman & Engel p. 614
 PDR pp. 1739, 1741
 OCDS Vol. 2 p. 110 (1980)
 DOT 15 (9) 411 (1979)
 I.N. p. 652
 REM p. 905
 Hauck, F.P., Cimarusti, C.M. and Narayanan, V.L.; U.S. Patent 3,935,267; January 27, 1976; assigned to E.R. Squibb & Sons, Inc.

NAFCILLIN SODIUM

Therapeutic Function: Antibacterial

Chemical Name: 6-(2-ethoxy-1-naphthamido)-3,3-dimethyl-7-oxo-4-thia-1-azabicyclo[3.2.0]-heptane-2-carboxylic acid sodium salt

Common Name: 6-(2-ethoxy-1-naphthamido)penicillin sodium salt

Structural Formula:

Chemical Abstracts Registry No.: 985-16-0; 147-52-4 (Base)

Trade Name	Manufacturer	Country	Year Introduced
Unipen	Wyeth	U.S.	1964
Nafcil	Bristol	U.S.	1976
Nallpen	Beecham	U.S.	1983
Naftopen	Gist-Brocades	—	—

Raw Materials

6-Aminopenicillanic acid
2-Ethoxy-1-naphthoyl chloride
Sodium bicarbonate

Manufacturing Process

A stirred suspension of 12.6 grams 6-aminopenicillanic acid in 130 ml dry alcohol-free chloroform was treated with 16 ml triethylamine and then with 13.8 grams of a solution of 2-ethoxy-1-naphthoyl chloride in 95 ml chloroform. After being washed successively with 58 ml each of 1N and then 0.1 N hydrochloric acid the chloroform solution was extracted with N aqueous sodium bicarbonate (58 ml + 6 ml). The combined bicarbonate extracts were washed with 20 ml ether and then evaporated at low temperature and pressure to give the crude sodium salt of 2-ethoxy-1-naphthylpenicillin [also called sodium 6-(2-ethoxy-1-naphthamido)penicillinate] as a yellow powder (20.3 grams). This was dissolved in 20 ml water at 30°C and diluted with 180 ml n-butanol, also at 30°C, with stirring. Slow cooling to 0°C gave colorless needles of the product.

References

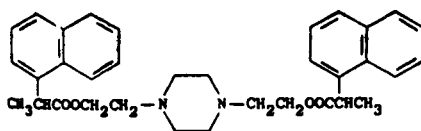
Merck Index 6199
Kleeman & Engel p. 615
PDR pp. 700, 1991
OCDS Vol. 1 p. 412 (1977)
I.N. p. 653
REM p. 1196
Doyle, F.P. and Nayler, J.H.C.; U.S. Patent 3,157,639; November 17, 1964; assigned to Beecham Group Limited, England

NAFIVERINE

Therapeutic Function: Antispasmodic

Chemical Name: α -methyl-1-naphthaleneacetic acid 1,4-piperazinediyl-di-2,1-ethanediy ester

Common Name: —

Structural Formula:**Chemical Abstracts Registry No.:** 5061-22-3

Trade Name	Manufacturer	Country	Year Introduced
Naftidan	De Angeli	Italy	1969

Raw Materials

α -Methyl-1-naphthylacetic acid
 Thionyl chloride
 N,N'-Di-(β -hydroxyethyl)piperazine

Manufacturing Process

15 grams of α -methyl-1-naphthylacetic acid were refluxed with 50 ml of thionyl chloride during 3 hours. The excess thionyl chloride was removed under reduced pressure and the product was also isolated by distillation under reduced pressure. Yield: 15.6 grams (96%). The α -methyl-1-naphthyl acetyl chloride boils at 120° to 124°C. 1.76 grams of N,N'-di-(β -hydroxyethyl)-piperazine, 1.9 grams of sodium bicarbonate and 4.45 grams of α -methyl-1-naphthyl acetyl chloride in 30 ml of anhydrous acetonitrile were refluxed with stirring during 5 hours. After cooling the mixture was filtered and the acetonitrile evaporated off under reduced pressure. 5.2 grams of crude ester were obtained. The hydrochloride, melting at 220° to 221°C, may be prepared by dissolving the ester in absolute ethanol and treating the solution with anhydrous gaseous hydrogen chloride.

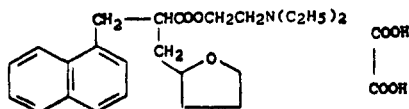
References

Merck Index 6200

I.N. p. 653

Pala, G.; British Patent 1,016,968; Jan. 12, 1966; assigned to Istituto de Angeli, SpA, Italy

NAFRONYL OXALATE

Therapeutic Function: Vasodilator**Chemical Name:** Tetrahydro- α -(1-naphthalenylmethyl)-2-furanpropanoic acid 2-(diethylamino)ethyl ester acid oxalate**Common Name:** Naftidofuryl**Structural Formula:****Chemical Abstracts Registry No.:** 3200-06-4; 31329-57-4 (Base)

Trade Name	Manufacturer	Country	Year Introduced
Dusodril	Roland	W. Germany	1968
Praxilene	Oberval	France	1968

Trade Name	Manufacturer	Country	Year Introduced
Prazilene	Lipha	U.K.	1972
Praxilene	Formenti	Italy	1973
Praxilene	Biochimica	Switz.	1980
Citoxid	Disprovent	Argentina	—

Raw Materials

β -(1-Naphthyl)- β' -tetrahydrofurfuryl isobutyric acid
 β -Chloroethyl-N-diethylamine
 Oxalic acid

Manufacturing Process

30 grams (0.106 mol) of β -(1-naphthyl)- β' -tetrahydrofurfuryl isobutyric acid are heated under reflux for 8½ hours in 230 cc of isopropanol with 14 grams (0.103 mol) of β -chloroethyl-N-diethylamine. After evaporation of the isopropanol in vacuo, the syrupy residue is treated with a solution of K_2CO_3 . Extraction with ether is carried out after drying over Na_2SO_4 .

Distillation of the extract yields 28.5 grams of a very viscous yellow liquid with a $BP_{0.95-1.09}$ millibar = 198° to 202°C. The yield is 70.5% (theoretical quantity = 40.5 grams). 1.3 grams (0.0103 mol) of dihydrated oxalic acid are dissolved while being made tepid in 8 cc of acetone. The cooled solution has added thereto 4 grams (0.0104 mol) of N-diethyl-aminoethyl- β -(1-naphthyl)- β' -tetrahydrofurfuryl isobutyrate, obtained according to the process described above and dissolved in 10 cc of acetone. The solution is brought to boiling point for 15 minutes. After cooling to ambient temperature, it is placed in a refrigerator. Crystallization occurs after 2 hours, the crystals which have formed are separated by centrifuging, and after washing in hexane and drying in vacuo 3.5 grams of white crystals are obtained. After being recrystallized three times, in alcohol and then in a mixture of alcohol and ethyl acetate, the product is analytically pure and has a MP = 110° to 111°C (heating stage).

References

Merck Index 6201

Kleeman & Engel p. 615

OCDS Vol. 2 p. 213 (1980)

DOT 5 (1) 19 (1969)

I.N. p. 654

Szarvasi, E. and Bayssat, M.; U.S. Patent 3,334,096; August 1, 1967; assigned to Lipha, Lyonnaise Industrielle Pharmaceutique, France

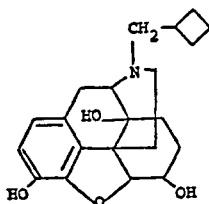
NALBUPHINE

Therapeutic Function: Analgesic

Chemical Name: N-cyclobutylmethyl-14-hydroxydihydronormorphinone

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 20594-83-6; 23277-43-2 (Hydrochloride)

Trade Name	Manufacturer	Country	Year Introduced
Nubain	Du Pont	U.S.	1979
Nubain	Du Pont	U.K.	1983

Raw Materials

14-Hydroxydihydronormorphinone
Cyclobutane carboxylic acid chloride
Lithium aluminum hydride

Manufacturing Process

To a slurry of 110.5 g of 14-hydroxydihydronormorphinone in 2.5 liters of methylene chloride and 280 ml of triethylamine was added a solution of 106 g of cyclobutanecarboxylic acid chloride in 500 ml of methylene chloride. The temperature of the reaction mixture was maintained at 20°C to 25°C during the addition. After 5 minutes the reaction mixture was brought to reflux and heated for 5 hours.

It was then cooled, washed with water, dried over sodium sulfate and evaporated to dryness. The residue was crystallized from benzene and pentane to give 138.5 g of the dicyclobutane-carbonyl derivative, melting point about 112°C (dec.).

The dicyclobutanecarbonyl derivative (136.7 g) was dissolved in 200 ml of tetrahydrofuran and added dropwise to a suspension of 34.2 g of lithium aluminum hydride in 1 liters of tetrahydrofuran. The temperature of the mixture rose to reflux during the addition. Reflux was maintained for 2 hours after the addition was completed. After cooling, 110 ml of ethyl acetate was added dropwise, followed by 30 ml of water, followed by a solution of 53 g of ammonium chloride in 125 ml of water. The resulting mixture was filtered and the inorganic precipitate was washed with methanol. Evaporation of the combined filtrates gave 66 g of N-cyclobutylmethyl-14-hydroxydihydronormorphinone, melting point 229°C to 231°C.

References

Merck Index 6203
DFU 2 (9) 613 (1977)
Kleeman & Engel p. 616
PDR p. 858
OCDS Vol. 2 p. 319 (1980)
DOT 16 (2) 51 (1980)
I.N. p. 654
REM p. 1109
Blumberg, H., Pachter, I.J. and Matossian, Z.; U.S. Patent 3,332,950; July 25, 1967; assigned to Endo Laboratories, Inc.

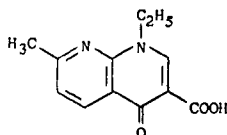
NALIDIXIC ACID

Therapeutic Function: Antibacterial

Chemical Name: 1-ethyl-1,4-dihydro-7-methyl-4-oxo-1,8-naphthyridine-3-carboxylic acid

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 389-08-2

Trade Name	Manufacturer	Country	Year Introduced
Neggram	Winthrop	U.S.	1964
Nalidixique	Winthrop	France	1974
Jicsron	Towa Yakuhin	Japan	1981
Baktogram	Farmakos	Yugoslavia	—
Betaxina	Amelix	Italy	—
Chemiurin	Cifa	Italy	—
Cybis	Breon	U.S.	—
Dixiben	Benvegna	Italy	—
Dixuroi	I.T.I.	Italy	—
Enexina	S.I.T.	Italy	—
Entolon	Sawai	Japan	—
Eucistin	San Carlo	Italy	—
Faril	Saita	Italy	—
Innoxalon	Sanko	Japan	—
Kusnarin	Kodama	Japan	—
Nali	Iltas	Turkey	—
Nalcidin	Schoum	Italy	—
Nalidicron	San-A	Japan	—
Nalidixico	Level	Spain	—
Nalidixin	Spofa	Czechoslovakia	—
Nalidixol	Hermes	Spain	—
Naligen	Sam	Italy	—
Naligram	Isis	Yugoslavia	—
Nalissina	Armour	Italy	—
Nalitucsan	Hishiyama	Japan	—
Nalix	Sigurta	Italy	—
Nalixan	Neofarma	Finland	—
Nalurin	Von Boch	Italy	—
Narigix	Taiyo	Japan	—
Naxuril	Esterfarm	Italy	—
Negabatt	Dessy	Italy	—
Nicelate	Toyo Jozo	Japan	—
Nogermin	Madaus	Spain	—
Notricel	Hortel	Spain	—
Pielos	S.T.I.P.	Italy	—
Poleon	Sumitomo	Japan	—
Renogram	Belupo	Yugoslavia	—
Restelon	Maruishi	Japan	—
Sicmylon	Niichiko	Japan	—
Specifin	Bergamon	Italy	—
Unaserus	Isei	Japan	—
Uralgin	Ceccarelli	Italy	—
Uretrene	Mitim	Italy	—
Uriben	R.P. Drugs	U.K.	—
Uriclar	Crosara	Italy	—
Uri-Flor	A.G.I.P.S.	Italy	—
Urigram	Trima	Israel	—
Urisco	I.C.I.	Italy	—
Uristeril	Ripari-Gero	Italy	—
Urodixin	Italchimici	Italy	—
Urogram	Firma	Italy	—
Urolex	Sirt-B.B.P.	Italy	—
Urolgin N	Takata	Japan	—
Uromina	Ausonia	Italy	—
Uroneg	Ibirn	Italy	—
Valuren	Intersint	Italy	—

Trade Name	Manufacturer	Country	Year Introduced
Wintomylon	Daiichi	Japan	—
Wintron	Tobishi	Japan	—

Raw Materials

2-Amino-6-methylpyridine
 Ethoxymethylenemalonic acid diethyl ester
 Sodium hydroxide
 Ethyl iodide

Manufacturing Process

A warm solution containing 41 grams of 4-hydroxy-7-methyl-1,8-naphthyridine-3-carboxylic acid and 39 grams of potassium hydroxide in 1 liter of ethanol and 200 cc of water was treated with 50 cc of ethyl iodide and the resulting mixture was refluxed gently overnight, acidified with hydrochloric acid and cooled. The resulting precipitate was collected and recrystallized twice from acetonitrile to yield 26 grams (56% yield) of 1-ethyl-7-methyl-4-oxo-1,8-naphthyridine-3-carboxylic acid, MP 229° to 230°C.

The starting material is prepared by reacting 2-amino-6-methylpyridine with ethoxymethylene-malonic acid diethyl ester and then reacting that product with sodium hydroxide.

References

Merck Index 6205
 Kleeman & Engel p. 616
 PDR p. 1922
 OCDS Vol. 1 p. 429 (1977) & 2, 370, 469 (1980)
 DOT 1 (1) 16 (1965)
 I.N. p. 33
 REM p. 1216
 Leshner, G.Y. and Gruett, M.D.; U.S. Patent 3,149,104; September 15, 1964; assigned to Sterling Drug Inc.

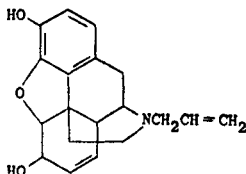
NALORPHINE

Therapeutic Function: Narcotic antagonist

Chemical Name: 7,8-didehydro-4,5-epoxy-17-(2-propenyl)morphinan-3,6-diol

Common Name: N-allylnormorphine

Structural Formula:



Chemical Abstracts Registry No.: 62-67-9; 57-29-4 (Hydrochloride)

Trade Name	Manufacturer	Country	Year Introduced
Nalline	MSD	U.S.	1952
Lethidrone	Wellcome	W. Germany	—
Nalorphine	Clin-Comar-Byla	France	—
Norfin	Lusofarmaco	Italy	—

Raw Materials

Normorphine
 Allyl bromide
 Sodium bicarbonate

Manufacturing Process

6 grams of normorphine, 2.7 grams of allyl bromide, 2.65 grams of sodium bicarbonate, and 75 cc of methanol were mixed together, and the resulting mixture was heated under reflux with stirring for a period of about 5½ hours. The reaction mixture was evaporated to dryness in vacuo, the residual material was extracted with 60 cc of boiling chloroform, 0.5 gram of activated charcoal was added, and the resulting mixture was filtered through a layer of diatomaceous silica. The filter cake was washed with four 10 cc portions of boiling chloroform, and the chloroform filtrate and washings were combined and evaporated to dryness in vacuo. The residual material was triturated with 25 cc of anhydrous ether until crystalline, the ethereal mixture was cooled, maintained at a temperature of 3°C overnight, filtered, and the crystalline mixture was washed with three 10 cc portions of ice-cold ether. The resulting crystalline product was dried to give 6.0 grams of N-allyl-normorphine, yield approximately 87% of theory, according to U.S. Patent 2,891,954.

References

Merck Index 6206

Kleeman & Engel p. 617

OCDS Vol. 1 p. 288 (1977) & 2, 318 (1980)

I.N. p. 655

REM p. 1106

Weijlard, J. and Erickson, A.E.; U.S. Patent 2,364,833; December 12, 1944; assigned to Merck & Co., Inc.

Weijlard, J.; U.S. Patent 2,891,954; June 23, 1959; assigned to Merck & Co., Inc.

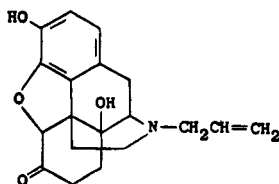
NALOXONE

Therapeutic Function: Narcotic antagonist

Chemical Name: 17-allyl-4,5 α -epoxy-3,14-dihydroxy-morphinan-6-one

Common Name: N-allylnoroxymorphone; N-allyl-1,4-hydroxydihydronormorphinone

Structural Formula:



Chemical Abstracts Registry No.: 465-65-6; 357-08-4 (Hydrochloride)

Trade Name	Manufacturer	Country	Year Introduced
Narcan	Du Pont	U.S.	1971
Narcan	Du Pont	U.K.	1975
Narcanti	Winthrop	W. Germany	1978
Narcan	Winthrop	France	1980
Narcan	Crinos	Italy	1980
Nalone	End	U.S.	—
Talwin	Winthrop-Breon	U.S.	—

Raw Materials

Oxymorphone	Acetic anhydride
Cyanogen bromide	Hydrogen chloride
Allyl bromide	

Manufacturing Process

10 grams of 14-hydroxydihydromorphinone (oxymorphone) was converted into its diacetate by warming it on the steam bath with 80 cc of acetic anhydride for about 2 hours. The acetic anhydride was removed on the water bath under a vacuum of about 30 mm absolute pressure. The melting point of the residue was 220°C. The residue was taken up in 100 cc of chloroform. An equal amount by weight of cyanogen bromide was added and the mixture was refluxed at about 60°C for about 5 hours. After refluxing, the mixture was washed with 100 cc of a 5% aqueous hydrochloric acid solution, dried over sodium sulfate and the chloroform removed by evaporation under a vacuum of about 30 mm. The residue had a melting point of 240°C.

The residue was then heated at about 90°C for 16 hours on a steam bath with 300 cc of 20% aqueous hydrochloric acid solution, and treated with a small amount, e.g., 1 gram of charcoal. The hydrochloric acid was then removed under a vacuum of 15 mm, the residue dissolved in 30 cc of water and precipitated by the addition of 2.4 cc of concentrated aqueous ammonia. The precipitate was filtered off and dried. It consists of 14-hydroxydihydromorphinone. It is soluble in ethanol.

The 14-hydroxydihydromorphinone was suspended in 200 cc of pure ethyl alcohol, half its weight of sodium bicarbonate and half its weight of allyl bromide added and the resulting mixture was refluxed at about 75°C for 48 hours. The solution was cooled, e.g., to 10°C and filtered and the alcohol removed under a vacuum of 30 mm. The residue was dissolved in chloroform and filtered. The chloroform was removed under a vacuum of 30 mm and the residue was crystallized from ethylacetate. The crystallized product, N-allyl-1,4-hydroxydihydromorphinone, has a melting point of 184°C, is soluble in chloroform and insoluble in petroleum ether. The yield amounts to 20% based on the weight of the reacted 14-hydroxydihydromorphinone.

References

- Merck Index 6208
- Kleeman & Engel p. 618
- PDR pp. 858, 1932
- OCDS Vol. 1 p. 289 (1977) & 2, 318, 323 (1980)
- DOT 8 (8) 295 (1972)
- I.N. p. 655
- REM p. 1106
- Lewenstein, M.J. and Fishman, J.; U.S. Patent 3,254,088; May 31, 1966

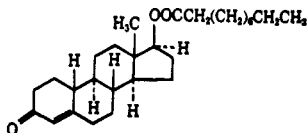
NANDROLONE DECANOATE

Therapeutic Function: Anabolic

Chemical Name: 17 β -[(1-oxodecyl)oxy] estr-4-en-3-one

Common Name: 19-nortestosterone decanoate; norandrostenolone decanoate

Structural Formula:



Chemical Abstracts Registry No.: 360-70-3; 434-22-0 (Base)

Trade Name	Manufacturer	Country	Year Introduced
Deca-Durabolin	Organon	U.S.	1962
Deca-Hybolin	Hyrex	U.S.	1979
Deca-Noralone	Taro	Israel	—
Fortabolin	Deva	Turkey	—
Iebolan	I.E. Kimya Evi	Turkey	—
Kabolin	Legere	U.S.	—
Methybol	Mepha	Switz.	—
Nordecon	Ibsa	Switz.	—
Sterobolin	Neofarma	Finland	—
Turinabol-Depot	Jenapharm	E. Germany	—

Raw Materials

19-Nortestosterone
Decanoic acid chloride

Manufacturing Process

1 gram of 19-nortestosterone is dissolved in 3 ml of dry pyridine, after which the resulting solution is cooled to -20°C . A solution of 1.0 gram of decanoic acid chloride in 3 ml of dry benzene is added to the cooled solution. The mixture is maintained at -15°C for 16 hours and then poured into ice water. The aqueous liquid is extracted with benzene, the benzene solution is washed with respectively 1 N sodium hydroxide solution, 2 N hydrochloric acid and with water until neutral reaction.

Then the solution is dried on sodium sulfate, filtered, and evaporated to dryness. The residue, 1.63 grams is dissolved in hexane, this solution is filtered over 30 grams of neutral aluminum oxide, and evaporated to dryness. On paper chromatographic investigation it turned out that the obtained 19-nortestosterone 17-decanoate which at room temperature is an oil consists of a single compound, according to U.S. Patent 2,998,423.

References

Merck Index 6212
Kleeman & Engel p. 620
PDR pp. 1033, 1286
OCDS Vol. 1 p. 171 (1977)
I.N. p. 655
REM p. 999

Donia, R.A. and Ott, A.C.; U.S. Patent 2,798,879; July 9, 1957; assigned to The Upjohn Company

De Wit, E.D. and Overbeek, G.A.; U.S. Patent 2,998,423; August 29, 1961; assigned to Organon Inc.

NANDROLONE PHENPROPIONATE

Therapeutic Function: Anabolic

Chemical Name: 17 β -hydroxyestr-4-en-3-one 3-phenylpropionate

Common Name: 19-nortestosterone β -phenylpropionate

Structural Formula: See Nandrolone Decanoate for the steroid structure

Chemical Abstracts Registry No.: 62-90-8; 434-22-0 (Base)

Trade Name	Manufacturer	Country	Year Introduced
Durabolin	Organon	U.S.	1959
Nandrolin	Tutag	U.S.	1979
Activin	Aristegui	Spain	—
Anticatabolin	Falorni	Italy	—
Hepa-Obaton	Nourypharma	W. Germany	—
Hybolin Improved	Hyrex	U.S.	—
Norabol	Pharmacia	Sweden	—
Noralone	Taro	Israel	—
Norandrol	Panther-Osfa	Italy	—
Norandros	Castillon	Spain	—
Norbalin	Bieffe	Italy	—
Noromon	Ibsa	Switz.	—
Norstenol	Ravizza	Italy	—
Sintabolin	A.F.I.	Italy	—
Strabolene	Isola-Ibi	Italy	—
Superanbolon	Spofa	Czechoslovakia	—
Superbolin	Labif	Italy	—
Turinabol	Jenapharm	E. Germany	—

Raw Materials

19-Nortestosterone
 β -Phenylpropionyl chloride

Manufacturing Process

An ice-cold solution of 1.5 grams of 19-nortestosterone and 1.5 ml of dry pyridine in 10 ml of dry benzene is prepared and a solution of 1.5 ml of β -phenylpropionyl chloride in 5 ml of dry benzene is added dropwise over a period of about 2 minutes with stirring. The resulting mixture is allowed to stand overnight under an atmosphere of nitrogen and then washed successively with cold 5% aqueous hydrochloric acid solution, cold 2.5% aqueous sodium hydroxide solution, and water. After drying over anhydrous sodium sulfate, the solvent is evaporated to give an almost colorless oil. Recrystallization from methanol gives white crystals of 19-nortestosterone 17- β -phenylpropionate, MP 91° to 92.5°C.

References

Merck Index 6214

Kleeman & Engel p. 621

PDR p. 1286

OCDS Vol. 1 p. 171 (1977)

I.N. p. 656

REM p. 999

Donia, R.A. and Ott, A.C.; U.S. Patent 2,868,809; January 13, 1959; assigned to The Upjohn Company

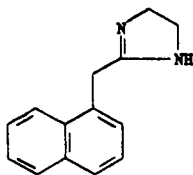
NAPHAZOLINE

Therapeutic Function: Nasal decongestant

Chemical Name: 4,5-dihydro-2-(1-naphthalenylmethyl)-1H-imidazole

Common Name: 2-(1-naphthylmethyl)imidazoline

Structural Formula:



Chemical Abstracts Registry No.: 835-31-4; 550-99-2 (Hydrochloride)

Trade Name	Manufacturer	Country	Year Introduced
Privine	Ciba	U.S.	1942
Albalon	Allergan	U.S.	1970
Naphcon Forte	Alcon	U.S.	1975
Clera	Person Covey	U.S.	1978
Vasoclear	Smith Miller & Patch	U.S.	1979
Opcon	Muro	U.S.	1981
Nafazair	Pharmafair	U.S.	1983
Actinophthyl	Gregoire	France	—
Bactio-Rhin	Byk Liprandi	Argentina	—
Biogan	Recip	Sweden	—
Coldan	Sigmapharm	Austria	—
Degest-2	Barnes-Hind	U.S.	—
Gotinal	Promeco	Argentina	—
Imidazyl	Tubi Lux Pharma	Italy	—
Imidin	Ysat Wernigerode	E. Germany	—
Imizol	Farmigea	Italy	—
Murine	Abbott	U.K.	—
Naftazolina	Bruschettini	Italy	—
Nafine	Ibsa	Switz.	—
Nasal Yer	Yer	Spain	—
Nomaze	Fisons	U.K.	—
Ocunasal	Sam-on	Israel	—
Pivanol	Tek	Turkey	—
Privin	Ciba	W. Germany	—
Proculin	Ankerwerk	E. Germany	—
Ran	Corvi	Italy	—

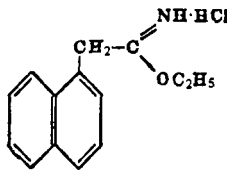
Trade Name	Manufacturer	Country	Year Introduced
Rhinex S	Ysat Wernigerode	E. Germany	—
Rhinon	Petrasch	Austria	—
Rimidol	Leo	Sweden	—
Rinofug	Chimimport Export	Rumania	—
Vasoconstrictor	Pensa	Spain	—
Vistalbalon	Pharm-Allergan	W. Germany	—

Raw Materials

Naphthyl-(1)-acetonitrile	Methanol
Ethanol	Ethylene diamine

Manufacturing Process

2.7 parts of naphthyl-(1)-acetiminoethylether hydrochloride of the formula



(produced from naphthyl-(1)-acetonitrile and methanol) are dissolved in 12 parts of absolute alcohol. 1 part of ethylenediamine is then added and the whole is heated to gentle boiling while passing nitrogen through it and simultaneously stirring until ammonia escapes no longer. The alcohol is then distilled and the residue mixed with 40 parts of benzene and 1.8 parts of caustic potash. Stirring is continued for some time whereby the imidazoline base is dissolved in benzene. The benzene residue is recrystallized several times from toluene. Reaction with HCl gives the hydrochloride.

References

- Merck Index 6218
- Kleeman & Engel p. 622
- PDR pp. 728, 809, 1549
- OCDS Vol. 1 p. 241 (1977)
- I.N. p. 657
- REM p. 888
- Sonn, A.; U.S. Patent 2,161,938; June 13, 1939; assigned to the Society of Chemical Industry in Basle, Switzerland

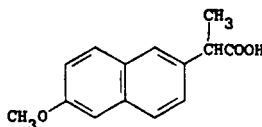
NAPROXEN

Therapeutic Function: Antiinflammatory

Chemical Name: (+)-6-methoxy- α -methyl-2-naphthaleneacetic acid

Common Name: d-2-(6-methoxy-2-naphthyl)propionic acid

Structural Formula:



Chemical Abstracts Registry No.: 22204-53-1

Trade Name	Manufacturer	Country	Year Introduced
Naprosyn	Syntex	U.K.	1973
Naprosyne	Cassenne	France	1975
Proxen	Gruenenthal	W. Germany	1975
Naprosyn	Recordati	Italy	1975
Naprosyn	Syntex	Switz.	1975
Naprosyn	Syntex	U.S.	1976
Naixan	Tanabe	Japan	1978
Congex	Nemi	Argentina	—
Floginax	Farmochimica	Italy	—
Gibixen	Gibipharm	Italy	—
Laser	Tosi-Novara	Italy	—
Madaprox	Madariaga	Spain	—
Naprium	Radium farma	Italy	—
Naprius	Magis	Italy	—
Naprox	Andromaco	Argentina	—
Naxyn	Teva	Israel	—
Novonaprox	Novopharm	Canada	—
Numide	Hosbon	Spain	—
Prexan	Lafare	Italy	—
Veradol	Schering	W. Germany	—
Xenar	Alfar Farma Clutici	Italy	—

Raw Materials

2-Bromo-6-methoxynaphthalene	Magnesium
Ethyl-2-bromopropionate	Cadmium chloride
Sodium hydroxide	

Manufacturing Process

According to U.S. Patent 3,658,858, a solution of 24 grams of 2-bromo-6-methoxynaphthalene in 300 ml of tetrahydrofuran is slowly added to 2.5 grams of magnesium turnings and 100 ml of tetrahydrofuran at reflux temperature. After the addition is complete, 20 grams of cadmium chloride is added, and the resultant mixture is refluxed for 10 minutes to yield a solution of di-(6-methoxy-2-naphthyl)cadmium (which can be separated by conventional chromatography, although separation is unnecessary).

A solution of 18 grams of ethyl 2-bromopropionate in 20 ml of tetrahydrofuran is then added to the cooled reaction mixture. After 24 hours at 20°C, the product is hydrolyzed by adding 200 ml of 5 weight percent methanolic sodium hydroxide followed by heating to reflux for 1 hour. The reaction mixture is then diluted with excess 1 N sulfuric acid and extracted with ether. The ether phase is separated, evaporated to dryness and the residue is recrystallized from acetone-hexane to yield 2-(6-methoxy-2-naphthyl)propionic acid.

References

- Merck Index 6269
 Kleeman & Engel p. 623
 PDR p. 1801
 OCDS Vol. 1 p. 86 (1977)
 DOT 9 (9) 384 (1973) & 10 (3) 95 (1974)
 I.N. p. 658
 REM p. 1119
 Alvarez, F.S.; U.S. Patent 3,637,767; January 25, 1972; assigned to Syntex Corp., Panama
 Harrison, I.T.; U.S. Patent 3,658,858; April 25, 1972; assigned to Syntex Corp., Panama
 Alvarez, F.S.; U.S. Patent 3,663,584; May 16, 1972; assigned to Syntex Corp., Panama

Alvarez, F.S.; U.S. Patent 3,694,476; September 26, 1972; assigned to Syntex Corp., Panama Halpern, O.; U.S. Patent 3,720,708; March 13, 1973

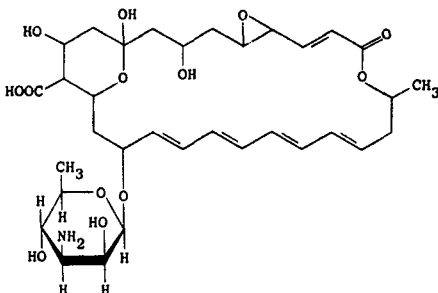
NATAMYCIN

Therapeutic Function: Antibacterial (ophthalmic)

Chemical Name: See Structural Formula

Common Name: Pimaricin

Structural Formula:



Chemical Abstracts Registry No.: 7681-93-8

Trade Name	Manufacturer	Country	Year Introduced
Pimafucine	Beytout	France	1964
Pimafucin	Brocades	U.K.	1965
Pimafucort	Brocades	Italy	1966
Pimafucin	Basotherm	W. Germany	1967
Natacyn	Alcon	U.S.	1979
Myprozine	Lederle	U.S.	—

Raw Materials

Bacterium *Streptomyces gilvosporeus*
Starch
Corn steep liquor

Manufacturing Process

The Fermentation Process: The process by which this antifungal substance is produced is an aerobic fermentation of an aqueous nutrient medium inoculated with a pimaricin-producing strain of *Streptomyces gilvosporeus*. The nutrient medium contains an assimilable source of carbon such as starch, molasses, or glycerol, an assimilable source of nitrogen such as corn steep liquor and inorganic cations such as potassium, sodium or calcium, and anions such as sulfate, phosphate or chloride. Trace elements such as boron, molybdenum or copper are supplied as needed in the form of impurities by the other constituents of the medium.

In more detail the nutrient medium used may contain sources of carbon such as starch, hydrolyzed starch, sugars such as lactose, maltose, dextrose, sucrose, or sugar sources such as molasses; alcohols, such as glycerol and mannitol; organic acids, such as citric acid and acetic acid; and various natural products which may contain other nutrient materials in addition to carbonaceous substances.

Nitrogen sources include proteins, such as casein, zein, lactalbumin; protein hydrolyzates such as proteoses, peptones, peptides, and commercially available materials, such as N-Z Amine which is understood to be a casein hydrolyzate; also corn steep liquor, soybean meal, gluten, cottonseed meal, fish meal, meat extracts, stick liquor, liver cake, yeast extracts and distillers' solubles; amino acids, urea, ammonium and nitrate salts. Such inorganic elements as sodium, potassium, calcium and magnesium; and chlorides, sulfates, phosphates and combinations of these anions and cations in the form of mineral salts may be advantageously used in the fermentation.

The so-called trace elements, such as boron, cobalt, iron, copper, zinc, manganese, chromium, molybdenum and still others may also be used to advantage. Generally, these trace elements occur in sufficient quantities in the carbonaceous and nitrogenous constituents of the medium, particularly if derived from natural sources, or in the tap water, and the addition of further quantities of these trace elements may consequently be unnecessary.

The fermentation liquor is aerated in the customary manner by forcing sterile air through the fermenting mixture usually at the rate of about 1 volume of air per volume of fermentation medium per minute. To minimize contamination with foreign microorganisms, the fermentation vessels should be closed and a pressure of 2 to 15 pounds above atmospheric pressure maintained in the vessel. In addition to the agitation provided by aeration, mechanical agitation is generally desirable. Antifoaming agents, such as 1% octadecanol in lard oil, may be added from time to time as required to prevent excessive foaming. Fermentation is conducted at a temperature preferably on the order of 26°C to 30°C but may be as low as 17°C or as high as 42°C.

The time required for maximum production of the antifungal substance will vary considerably depending upon other conditions of the fermentation. Generally, about 48 hours is required before appreciable quantities of the antifungal substance are detected in the medium. The production of the antifungal substance increases with time, and the fermentation may run as long as 120 hours. The hydrogen ion conditions normally vary from about pH 6 to pH 8.0, although deviations from these values are permissible, according to British Patent 846,933. The reader is referred to the patents cited for details of pimarinic purification.

References

Merck Index 6278

Kleeman & Engel p. 624

DOT 14 (6) 255 (1978)

I.N. p. 659

REM p. 1230

Koninklijke Nederlandsche Gist- & Spiritusfabriek N.V., Netherlands; British Patent 844,289; August 10, 1960

American Cyanamid Company; British Patent 846,933; September 7, 1960

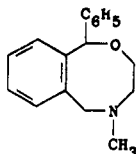
NEFOPAM HYDROCHLORIDE

Therapeutic Function: Muscle relaxant; antidepressant

Chemical Name: 3,4,5,6-tetrahydro-5-methyl-1-phenyl-1H-2,5-benzoxazine hydrochloride

Common Name: —

Structural Formula:



(base)

Chemical Abstracts Registry No.: 13669-70-0 (Base)

Trade Name	Manufacturer	Country	Year Introduced
Ajan	Kettelnack	W. Germany	1976
Acupan	Carnegie	U.K.	1978
Acupan	Riker	France	1981
Lenipan	Chiesi	Italy	1981
Oxadol	I.S.I.	Italy	1982
Acupan	Boehr. Mann.	Italy	1983

Raw Materials

2-Benzoylbenzoic acid	Thionyl chloride
2-Methylaminoethanol	Lithium aluminum hydride
p-Toluenesulfonic acid	Hydrogen chloride

Manufacturing Process

The starting material is prepared by reacting 2-benzoylbenzoic acid with thionyl chloride and then with 2-methylaminoethanol. 20.0 grams (0.07 mol) of N-(2-hydroxyethyl)-N-methyl-p-benzoylbenzamide is suspended in 100 ml tetrahydrofuran and then slowly added in small portions to a solution of 5.5 grams (0.14 mol) of lithium aluminum hydride in 150 ml tetrahydrofuran with cooling and stirring. The mixture is then refluxed for 18 hours, cooled and then to it is successively added 5.5 ml water, 5.5 ml of 3.75 N sodium hydroxide and 16 ml water. After removal of precipitated salts by filtration, the solution remaining is concentrated under reduced pressure and the residue dried to yield 19.5 grams of crude product. Yield after conversion to the hydrochloride salt and recrystallization is 17.0 grams (89%), MP 128° to 133°C.

5-methyl-1-phenyl-1,3,4,6-tetrahydro-5H-benz[f]-2,5-oxazocine is prepared as follows. 3.0 grams (0.011 mol) of 2-([N-(2-hydroxyethyl)-N-methyl]amino)methylbenzhydrol, prepared as described above, 3.0 grams p-toluenesulfonic acid and 15 ml benzene are heated together with stirring until all the benzene is distilled off. The residual oil is heated to 105°C and held at this temperature for 1 hour, then cooled and dissolved in 30 ml water. This aqueous solution is then basified to pH 10.0 with 12 N sodium hydroxide, extracted with ether, and the extracts washed with water, dried over anhydrous sodium sulfate and the solvent removed under reduced pressure. The 2.26 grams (81%) oil remaining is converted to the hydrochloride salt, MP 238° to 242°C.

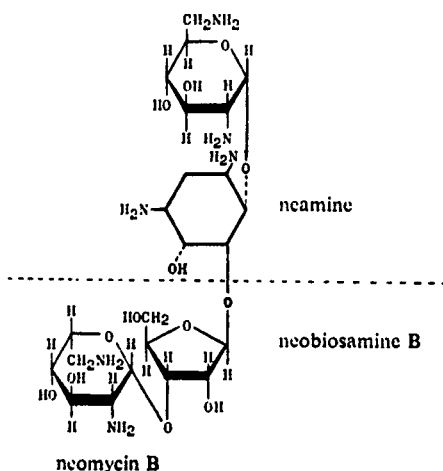
References

- Merck Index 6287
- Kleeman & Engel p. 626
- OCDS Vol. 2 p. 447 (1980)
- DOT 12 (7) 275 (1976)
- I.N. p. 661
- Baltes, B.J.; U.S. Patent 3,487,153; December 30, 1969; assigned to Rexall Drug and Chemical Company

NEOMYCIN

Therapeutic Function: Antibacterial

Chemical Name: O-2,6-diamino-2,6-dideoxy- α -D-glucopyranosyl-(1 \rightarrow 3)-O- β -D-ribofuranosyl-(1 \rightarrow 5)-O-[2,6-diamino-2,6-dideoxy- α -D-glucopyranosyl-(1 \rightarrow 4)]-2-deoxy-D-streptomine

Common Name: Framycetin**Structural Formula:****Chemical Abstracts Registry No.:** 1404-04-2; 4146-30-9 (Sulfate)

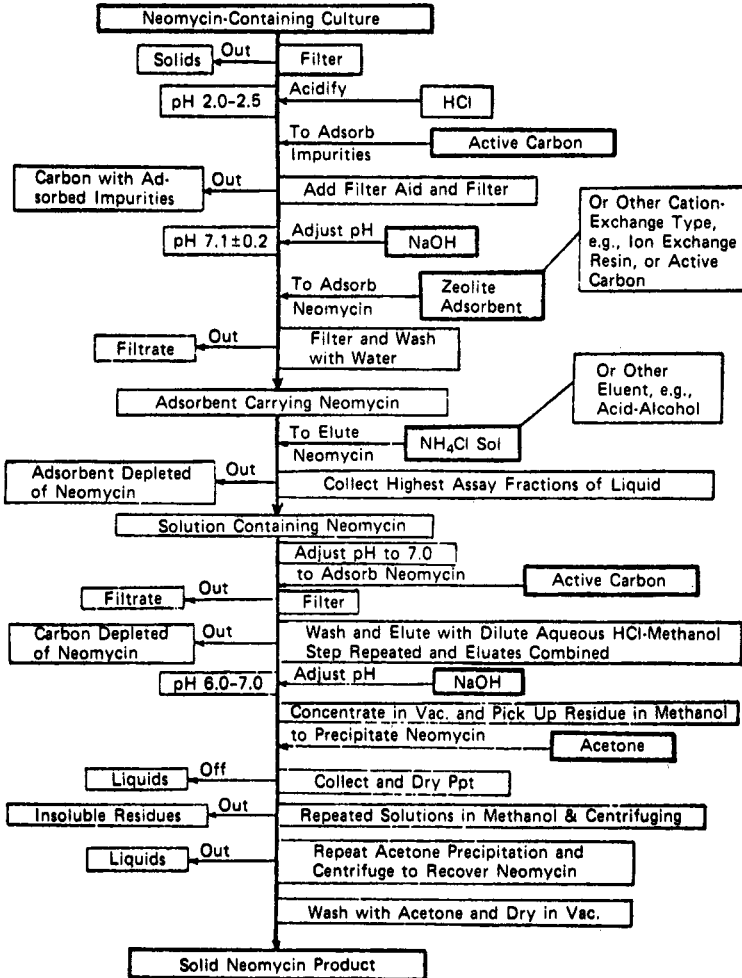
Trade Name	Manufacturer	Country	Year Introduced
Myciguent	Upjohn	U.S.	1951
Otobiotic	Schering	U.S.	1954
Mycifradin	Upjohn	U.S.	1957
Neobiotic	Pfizer	U.S.	1958
Apokalin	A.L.	Norway	—
Biofradin	Uriach	Spain	—
Bykomycin	Byk-Gulden	W. Germany	—
Cortisporin	Burroughs-Wellcome	U.S.	—
Dexmy	Takeda	Japan	—
Endomixin	Lusofarmaco	Italy	—
Fradio	Nippon Kayaku	Japan	—
Fradyl	Christiaens	Belgium	—
Ivax	Boots	U.K.	—
Larmicin	Larma	Spain	—
Myacyne	Werner Schnur	W. Germany	—
Mytrex	Savage	U.S.	—
Neobretin	Norbrook	U.K.	—
Neodecadron	MSD	U.S.	—
Neointestin	Hosbon	Spain	—
Neolate	Therafarm	U.K.	—
Neomicina Roger	Roger	Spain	—
Neomin	Glaxo	U.K.	—
Neo-Polycin	Merrell Dow	U.S.	—
Neopt	Sigma	Australia	—
Neosporin	Burroughs-Wellcome	U.S.	—
Neosulf	Protea	Australia	—
Neo-Synalar	Syntex	U.S.	—
Octicair	Pharmafair	U.S.	—
Otocort	Lemmon	U.S.	—
Siquent	Sigma	Australia	—
Tampovagan	Norgine	U.K.	—
Topisporin	Pharmafair	U.S.	—
Tri-Thalmic	Schein	U.S.	—

Raw Materials

Bacterium *Streptomyces fradiae*
Nutrient medium

Manufacturing Process

Neomycin has been produced by growing the organism, *Streptomyces* No. 3535, in a suitable nutrient medium under appropriate stationary or submerged aerobic (viz shaken) conditions, and then isolating and purifying the substance, e.g., by procedure of the sort described in the figure including various steps of adsorption, recovery by elution, separation from impurities, and precipitation.



Neomycin is usually used as the sulfate.

References

Merck Index 6300

Kleeman & Engel 626

PDR pp. 673, 738, 756, 888, 993, 1034, 1206, 1232, 1429, 1569, 1604, 1800

I.N. p. 663

REM p. 1181

Waksman, S.A. and Lechevalier, H.A.; U.S. Patent 2,799,620; July 16, 1957; assigned to

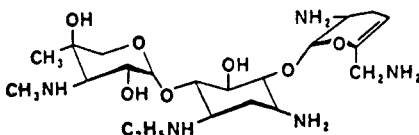
Rutgers Research and Educational Foundation

Jackson, W.G.; U.S. Patent 2,848,365; August 19, 1958; assigned to The Upjohn Company

Miller, T.W.; U.S. Patent 3,005,815; October 24, 1961; assigned to Merck & Co., Inc.

Moses, W.; U.S. Patent 3,022,228; February 20, 1962; assigned to S.B. Penick & Company

Haak, W.J.; U.S. Patent 3,108,996; October 29, 1963; assigned to The Upjohn Company

NETILMICIN**Therapeutic Function:** Antibiotic**Chemical Name:** 1-N-Ethylisomicin**Common Name:** —**Structural Formula:****Chemical Abstracts Registry No.:** 56391-56-1

Trade Name	Manufacturer	Country	Year Introduced
Netromycine	Schering	Switz.	1980
Certomycin	Byk-Essex	W. Germany	1980
Netillin	Kirby-Warrick	U.K.	1981
Netromicine	Unicet	France	1981
Nettacin	Essex	Italy	1982
Netromycin	Schering	U.S.	1983

Raw Materials

Sisomicin	Sulfuric acid
Acetaldehyde	Sodium cyanoborohydride

Manufacturing Process

To a solution of 5 g of sisomicin in 250 ml of water add 1 N sulfuric acid until the pH of the solution is adjusted to about 5. To the solution of sisomicin sulfuric acid addition salt thereby formed, add 2 ml of acetaldehyde, stir for 10 minutes, then add 0.85 g of sodium cyanoborohydride. Continue stirring at room temperature for 15 minutes, then concentrate solution in vacuo to a volume of about 100 ml, treat the solution with a basic ion exchange resin [e.g., Amberlite IRA 401S (OH⁻)], then lyophilize to a residue comprising 1-N-ethyl-sisomicin.

Purify by chromatographing on 200 g of silica gel, eluting with lower phase of a chloroform-methanol-7% aqueous ammonium hydroxide (2:1:1) system. Combine the eluates as deter-

mined by thin layer chromatography and concentrate the combined eluates of the major component in vacuo to a residue comprising 1-N-ethylisonicotin (yield 1.25 g). Further purify by again chromatographing on 100 g of silica gel eluting with a chloroform-methanol-3.5% ammonium hydroxide (1:2:1) system. Pass the combined, like eluates (as determined by thin layer chromatography) through a column of basic ion exchange resin and lyophilize the eluate to obtain 1-N-ethylisonicotin (yield 0.54 g).

There is also a fermentation route to netilmicin as noted by Kleeman & Engel.

References

Merck Index 6322

DFU 3 (7) 527 (1978)

Kleeman & Engel p. 627

PDR p. 1635

DOT 17 (8) 324 (1981)

I.N. p. 666

REM p. 1183

Wright, J.J., Daniels, P.J.L., Mallams, A.K. and Nagabhushan, T.L.; U.S. Patent 4,002,742; January 11, 1977; assigned to Schering Corp.

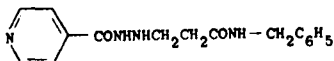
NIALAMIDE

Therapeutic Function: Antidepressant

Chemical Name: 4-Pyridinecarboxylic acid 2-[3-oxo-3-[(phenylmethyl)-amino] propyl] hydrazide

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 51-12-7

Trade Name	Manufacturer	Country	Year Introduced
Niamid	Pfizer	U.S.	1959
Niamide	Pfizer	France	1960
Niamid	Taito-Pfizer	Japan	—
Nuredal	Egyt	Hungary	—
SurgeX	Firma	Italy	—

Raw Materials

Isoniazid

Methyl acrylate

Benzylamine

Manufacturing Process

Methyl acrylate, 28.0 g (0.4 mol) was added dropwise during one hour to a solution containing 54.8 g (0.4 mol) of isonicotinic acid hydrazide (isoniazid) and 10 ml of glacial acetic acid in 400 ml of tertiary butyl alcohol. The resulting solution then was heated for 18 hours on a steam bath. Concentration of the reaction mixture to 100 ml yielded 13.0 g of unreacted isonicotinic acid hydrazide. The filtrate was concentrated to a thick syrup which was triturated

with anhydrous ether and recrystallized from isopropyl alcohol; MP 87°C to 88.5°C. Elemental analysis of the product gave 1-isonicotinyl-2-(β -carbomethoxyethyl)hydrazine.

A slurry of 7.5 g (0.034 mol) of 1-isonicotinyl-2-(carbomethoxyethyl)-hydrazine and 5 ml of benzylamine is heated with stirring at 130°C for three hours. The cooled mass is then recrystallized from ethyl acetate to yield white needles melting at 151.1°C to 152.1°C.

References

Merck Index 6330

Kleeman & Engel p. 628

OCDs Vol. 1 p. 254 (1977)

I.N. p. 667

Bloom, B.M. and Carnahan, R.E.; U.S. Patent 2,894,972; July 14, 1959; assigned to Chas. Pfizer & Co., Inc.

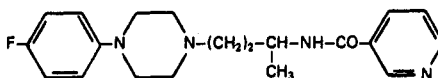
NIAPRAZINE

Therapeutic Function: Antihistamine

Chemical Name: 1-(4-Fluorophenyl)-4-[3-(3-pyridoyl)amino] butyl-piperazine

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 27367-90-4

Trade Name	Manufacturer	Country	Year Introduced
Nopron	Carrion	France	1976
Norpron	Riom	Italy	—

Raw Materials

1-(4-Fluorophenyl)piperazine dihydrochloride
 Trioxymethylene
 Acetone
 Hydroxylamine hydrochloride
 Lithium aluminum hydride
 Nicotinic acid chloride

Manufacturing Process

1st Stage: 10 ml of concentrated (10N) hydrochloric acid and 240 ml of acetone were added to a solution of 217.5 g (1 mol) of 1-(4-fluorophenyl)piperazine dihydrochloride in 400 ml of 96% ethanol. 50 g of powdered trioxymethylene were then added and the mixture was then slowly heated to reflux, which was maintained for 1 hour. A further 60 g of trioxymethylene were then added and heating to reflux was continued for a further 6 hours.

The mixture was then cooled, the precipitate formed was filtered off, washed with acetone and recrystallized from 96% ethanol.

The base was liberated from its salt by taking up the product in an aqueous solution of

sodium bicarbonate. The precipitate of the base thus obtained was recrystallized from petroleum ether to give 160 g of the desired product; melting point 46°C; yield 64%.

2nd Stage: 45.5 g (0.65 mol) of hydroxylamine hydrochloride were added to a solution of 128 g (0.5 mol) of the amino-ketone obtained in the preceding stage in 100 ml of ethanol and 40 ml of water. The mixture was allowed to react for 15 minutes at room temperature and was then heated to reflux for ½ hour. A part of the solvent was then distilled off and the product was then allowed to crystallize on cooling. After recrystallization from 96% ethanol, 117 g of the desired product were obtained; melting point 170°C; yield 77%.

3rd Stage: 93 g (0.35 mol) of the oxime obtained in the preceding stage, in the form of the base, were added in portions to a suspension of 17 g (0.45 mol) of lithium aluminum hydride in 400 ml of anhydrous ether. The mixture was then heated to reflux for 15 hours.

10 ml of ethyl acetate and then 50 ml of dilute caustic soda were added slowly with the usual precautions to the mixture. The organic phase was separated, dried over anhydrous Na₂SO₄, the solvent was distilled off and the residue obtained was distilled under reduced pressure to give 51 g of a thick oil; boiling point (2 mm Hg), 142°C to 143°C; yield 58%.

4th Stage: 10 ml of triethylamine were added in a solution of 25.2 g (0.1 mol) of the amine obtained in the preceding stage in 100 ml of anhydrous chloroform and the mixture was cooled to 2°C to 3°C. While maintaining this temperature, 17 g (0.12 mol) of nicotinic acid chloride were added with vigorous agitation.

After evaporation of the solvent, the residue was washed with water, the product taking the form of a mass. After recrystallization from ethyl acetate, a constant melting point of 131°C was obtained.

References

Merck Index 6331

Kleeman & Engel p. 628

DOT 13 (1) 29 (1977)

I.N. p. 667

Mauvernay, R.Y., Busch, N., Simond, J. and Moleyre, J.; U.S. Patent 3,712,893; January 23, 1973; assigned to SA Centre Europeen De Recherches Mauvernay, CERM

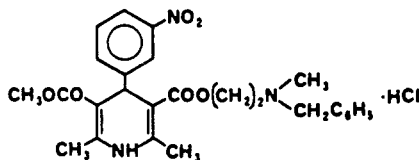
NICARDIPINE

Therapeutic Function: Cerebral vasodilator

Chemical Name: 2,6-Dimethyl-4-(3-nitrophenyl)-3-methoxycarbonyl-1,4-dihydropyridine-5-carboxylic acid-2-(N-benzyl-N-methylamino)ethyl ester hydrochloride

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 55985-32-5

Trade Name	Manufacturer	Country	Year Introduced
Nicodel	Mitsui	Japan	1981
Perdipin	Yamanouchi	Japan	1981

Raw Materials

Acetoacetic acid N-benzyl-N-methylaminoethyl ester
 β -Aminocrotonic acid methyl ester
 m-Nitrobenzaldehyde

Manufacturing Process

A mixture of 4.98 g of acetoacetic acid N-benzyl-N-methylaminoethyl ester, 2.3 g of β -aminocrotonic acid methyl ester, and 3 g of m-nitrobenzaldehyde was stirred for 6 hours at 100°C in an oil bath. The reaction mixture was subjected to a silica gel column chromatography (diameter 4 cm and height 25 cm) and then eluted with a 20:1 mixture of chloroform and acetone. The effluent containing the subject product was concentrated and checked by thin layer chromatography. The powdery product thus obtained was dissolved in acetone and after adjusting the solution with an ethanol solution saturated with hydrogen chloride to pH 1-2, the solution was concentrated to provide 2 g of 2,6-dimethyl-4-(3'-nitrophenyl)-1,4-dihydropyridine-3,5-dicarboxylic acid 3-methylester-5- β -(N-benzyl-N-methylamino)ethyl ester hydrochloride. The product thus obtained was then crystallized from an acetone mixture, melting point 136°C to 140°C (decomposed).

References

Merck Index 6334

DFU 2 (6) 409 (1977) (as Yc-93) & 4 (12) 911 (1979)

OCDS Vol. 3 p. 150 (1984)

DOT 18 (7) 325 (1982)

I.N. p. 668

Murakami, M., Takahashi, K., Iwanami, M., Fujimoto, M., Shibanuma, T., Kawai, R. and Takenaka, T.: U.S. Patent 3,985,758; October 12, 1976; assigned to Yamanouchi Pharmaceutical Co., Ltd.

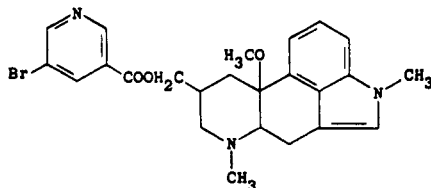
NICERGOLINE

Therapeutic Function: Peripheral vasodilator

Chemical Name: 10-methoxy-1,6-dimethylergoline-8 β -methanol 5-bromonicotinate (ester)

Common Name: Nicotergoline; 1-methyllumilysergol-8-(5-bromonicotinate) 10-methyl ether

Structural Formula:



Chemical Abstracts Registry No.: 27848-84-6

Trade Name	Manufacturer	Country	Year Introduced
Sermion	Farmitalia	Italy	1974
Sermion	Specia	France	1975
Nicergolyn	Farnex	Italy	—
Nicotergoline	Carlo Erba	Italy	—
Varson	Almirall	Spain	—
Vasospan	Exa	Argentina	—

Raw Materials

1-Methyl-lumilysergic acid	Methanol
Lithium aluminum hydride	Hydrogen chloride
5-Bromonicotinyl chloride	

Manufacturing Process

Preparation of 1-Methyl Lumilysergic Acid 8-Methyl Ester-10-Methyl Ether: Into a suspension of 10 grams of 1-methyl-lumilysergic acid in 600 cc of absolute methanol a stream of anhydrous hydrogen chloride is bubbled for 1.5 hours with strong cooling. The stream of hydrogen chloride is stopped and the mixture is allowed to stand for 30 minutes at 0°C, and is evaporated in vacuo to dryness. The residue is taken up with ice-cooled water made alkaline with concentrated ammonia and extracted with chloroform. The combined chloroform extracts are washed first with a 5% aqueous solution of sodium bicarbonate, then with water, and are thereafter dried over anhydrous sodium sulfate and finally evaporated in vacuo to dryness.

Preparation of 1-Methyl Lumilysergol-10-Methyl Ether: To a boiling suspension of 2 grams of lithium aluminum hydride in 50 cc of anhydrous tetrahydrofuran, a solution of 1 gram of 1-methyl lumilysergic acid-8-methyl ester-10-methyl ether in 20 cc of anhydrous tetrahydrofuran is added dropwise and the resulting solution is refluxed for a further 2 hours. After cooling the resulting solution, aqueous tetrahydrofuran is added to destroy the excess reducing agent and the solution is filtered. Tetrahydrofuran is distilled off and the residue is recrystallized from acetone petroleum ether.

Preparation of Nicergoline: To a solution of 1-methyl lumilysergol-10-methyl ether in pyridine, 5-bromonicotinyl chloride is used as an acylating agent at room temperature. The mixture is stirred for 1 hour. Water and methanol are added and the resulting mixture is stirred for 1 hour, extracted with chloroform, and washed in sequence with 1% aqueous caustic soda, 5% aqueous sodium bicarbonate solution, and water. The resulting solution is dried over anhydrous sodium sulfate and the solvent is distilled off. By recrystallization of the residue from acetone petroleum ether, nicergoline is obtained, melting at 136° to 138°C.

References

- Merck Index 6335
- Kleeman & Engel p. 629
- OCDs Vol. 2 p. 478 (1980)
- DOT 10 (12) 342 (1974)
- I.N. p. 668
- Bernardi, L., Bosisio, G. and Goffredo, O.; U.S. Patent 3,228,943; January 11, 1966; assigned to Società Farmaceutici Italia, Italy

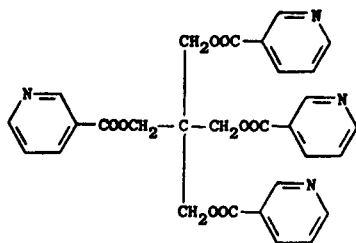
NICERITROL

Therapeutic Function: Cholesterol reducing agent

Chemical Name: 3-pyridinecarboxylic acid 2,2-bis[[(3-pyridinylcarbonyl)oxy] methyl]-1,3-propanediyl ester

Common Name: Pentaerythritol tetranicotinate

Structural Formula:



Chemical Abstracts Registry No.: 5868-05-3

Trade Name	Manufacturer	Country	Year Introduced
Cardiolipol	Gremy/Longuet	France	1972
Percyt	Sanwa	Japan	1979
Percyt	Tosi	Italy	1980
Percyt	Astra	Sweden	—

Raw Materials

Nicotinic acid chloride
Pentaerythritol
Pyridine

Manufacturing Process

160 grams of nicotinic acid chloride is charged into and made to react with 35 grams of pentaerythritol dissolved in 600 grams of dried, stabilized chloroform and 100 grams of carefully dried pyridine. Pyridinehydrochloride, pyridine and the excess of nicotinic acid chloride are removed through repeated extraction with water at a pH of approximately 3. Pentaerythritol nicotinate remains in the chloroform phase and is extracted by forming the hydrochloric acid salt of the ester using 1,000 ml of aqueous HCl at a pH of 1. The strongly acid extract is thereafter extracted several times with toluene. The acid extract is allowed to stand at room temperature for several hours in the presence of active carbon and the substance known as Versenate, i.e., the disodium salt of ethylene diamine tetraacetic acid; it is then filtered and pentaerythritol nicotinate is precipitated as a white, amorphous substance using 25% w/v aqueous ammonia, while stirring. Recrystallization of the product from ethyl alcohol gives flaky crystals, according to British Patent 1,022,880.

References

Merck Index 6336
Kleeman & Engel p. 630
I.N. p. 668
AB Bofors, Sweden; British Patent 1,022,880; March 16, 1966
AB Bofors, Sweden; British Patent 1,053,689; January 4, 1967

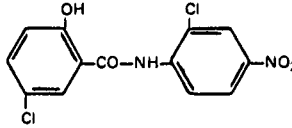
NICLOSAMIDE

Therapeutic Function: Anthelmintic

Chemical Name: 2',5-Dichloro-4'-nitrosalicylanilide

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 50-65-7

Trade Name	Manufacturer	Country	Year Introduced
Yomesan	Bayer	W. Germany	1960
Yomesan	Bayer	U.K.	1961
Yomesan	Bayer	Italy	1962
Tredemine	Roger Bellon	France	1964
Niclocide	Miles	U.S.	1982
Anti-Tenia	Uranium	Turkey	—
Atenase	I.C.N.-Usafarma	Brazil	—
Radeverm	Arzneimittelwerk Dresden	E. Germany	—
Teniarene	A.M.S.A.	Italy	—
Tenisid	Liba	Turkey	—

Raw Materials

5-Chlorosalicylic acid
2-Chloro-4-nitroaniline
Phosphorus trichloride

Manufacturing Process

17.2 g of 5-chlorosalicylic acid and 20.8 g of 2-chloro-4-nitroaniline are dissolved in 250 ml of xylene. While boiling, there are introduced slowly 5 g of PCl_3 . Heating is continued for 3 further hours. The mixture is then allowed to cool down and the crystals which separate are filtered off with suction. The crude product may be recrystallized from ethanol, melting at 233°C .

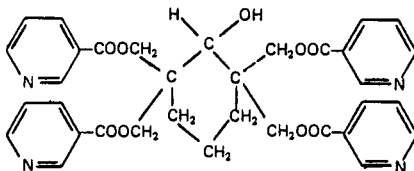
References

Merck Index 6356
Kleeman & Engel p. 630
PDR p. 1260
OCDS Vol. 2 p. 94 (1980)
I.N. p. 669
REM p. 1236
Schraufstatter, E. and Gonnert, R.; U.S. Patent 3,147,300; September 1, 1964; assigned to Farbenfabriken Bayer A.G.

NICOMOL

Therapeutic Function: Anticholesterol

Chemical Name: Cyclohexanol-2,2,6,6-tetrakis(hydroxymethyl)tetratricotinate

Common Name: —**Structural Formula:****Chemical Abstracts Registry No.:** 27959-26-8

Trade Name	Manufacturer	Country	Year Introduced
Cholexamine	Kyorin	Japan	1971
Acenol	Kissei	Japan	1981
Nicolanta	Sawai	Japan	—

Raw Materials

2,2,6,6-Tetramethylcyclohexanol
Nicotinic acid chloride

Manufacturing Process

To a mixture of 60 cc of benzene, 40 cc of pyridine and 17 g of hydrochloric acid salt of nicotinic acid chloride, was added 4.5 g of 2,2,6,6-tetramethylcyclohexanol, and the whole mixture was refluxed at 75°C to 80°C for 2.5 hours. After the mixture was cooled water was added. Precipitate formed was separated by filtration, washed thoroughly with water and dried. Recrystallization from dilute acetic acid gave 14 g of the final compound, melting point 177°C to 180°C.

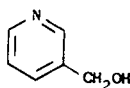
References

Merck Index 6360

DOT 7 (5) 173 (1971)

I.N. p. 670

Irikura, T., Sato, S., Abe, Y. and Kasuga, K.; U.S. Patent 3,299,077; January 17, 1967; assigned to Kyorin Seiyaku KK

NICOTINYL ALCOHOL**Therapeutic Function:** Peripheral vasodilator**Chemical Name:** 3-pyridinemethanol**Common Name:** 3-pyridylcarbinol**Structural Formula:****Chemical Abstracts Registry No.:** 100-55-0

Trade Name	Manufacturer	Country	Year Introduced
Roniacol	Roche	U.S.	1949
Danaden	Cascan	W. Germany	—
Peritard	Ikapharm	Israel	—
Ronicol	Roche	U.K.	—
Thilocombin	Thilo	W. Germany	—

Raw Materials

3-Cyanopyridine	Hydrogen
Ethyl alcohol	Nitrosyl chloride

Manufacturing Process

The catalyst is prepared by suspending 5 kg of catalyst grade charcoal in 200 liters of water, in a pressure vessel, and adding thereto 25 liters of 4% (as Pd metal) aqueous palladous chloride. Air is displaced from the vessel and then hydrogen is passed into the aqueous mixture at a pressure of 3 to 5 psi, while stirring, until no further absorption is noted and the chloride is completely reduced to metal.

To the aqueous suspension of the palladized charcoal catalyst thus obtained are added 20.8 kg of 3-cyano-pyridine (96% purity); and then are added 70 liters of a hydrochloric acid solution prepared by diluting 30 liters of 36% HCl with 40 liters of water. This represents approximately 1.75 mols of HCl for each mol of 3-cyano-pyridine. The suspension is maintained at 10° to 15°C and stirred continuously while introducing a current of hydrogen at a pressure of 3 to 5 psi. When absorption of hydrogen ceases and the 3-cyano-pyridine is completely reduced, the reaction mixture is filtered to remove the catalyst. The filter cake is washed with 40 liters of water in two equal portions, and the wash water is added to the filtrate.

The combined liquors, which comprise an aqueous hydrochloric acid solution of 3-amino-methyl-pyridine hydrochloride, are then heated to a temperature of 60° to 65°C, and ethyl nitrite gas is passed into the heated solution. The ethyl nitrite is generated by placing 20 liters of 90% ethyl alcohol in a suitable vessel, diluting with 200 liters of water, and, while stirring, adding to the dilute alcohol 18.3 kg of nitrosyl chloride at the rate of 2.25 kg per hour. (The process using methyl nitrite is carried out by substituting a stoichiometrically equivalent quantity of methyl alcohol for the ethyl alcohol.)

When all the ethyl nitrite has been added, the reaction mixture is refluxed for approximately one hour, then concentrated to dryness under reduced pressure (25 to 30 mm Hg) and at a maximum temperature of 70°C. The crystalline residue is dissolved in 35 liters of water and adjusted to a pH of 8 to 9 by addition (with cooling and stirring) of 11 to 12 kg of caustic soda. The sodium chloride formed is filtered off, and the filter cake is washed with 20 liters of normal butyl alcohol. This wash liquid is used for the first extraction of the product from the aqueous filtrate. The filtrate is then further extracted with four successive 20-liter portions of n-butyl alcohol.

All the extracts are combined and concentrated in vacuo (100°C/20 mm) to remove the n-butyl alcohol. The residue is submitted to fractionation under reduced pressure. The forerun (up to 112°C/2 to 3 mm) consists of a small amount of n-butyl alcohol and some 3-pyridylcarbinol. The main fraction, boiling at 112° to 114°C/2 to 3 mm, consists of 3-pyridylcarbinol.

References

- Merck Index 6369
- Kleeman & Engel p. 633
- I.N. p. 672
- REM p. 852

Ruzicka, L. and Prelog, V.; U.S. Patent 2,509,171; May 23, 1950; assigned to Ciba Limited, Switzerland

Cohen, A.; U.S. Patent 2,520,037; August 22, 1950; assigned to Hoffmann-La Roche Inc.

Schlöpfer, R.; U.S. Patent 2,547,048; April 3, 1951; assigned to Hoffmann-La Roche Inc.

Chase, G.O.; U.S. Patent 2,615,896; October 28, 1952; assigned to Hoffmann-La Roche Inc.

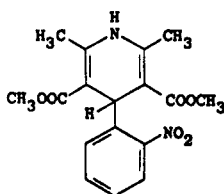
NIFEDIPINE

Therapeutic Function: Coronary vasodilator

Chemical Name: 1,4-dihydro-2,6-dimethyl-4-(2'-nitrophenyl)-3,5-pyridinedicarboxylic acid dimethyl ester

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 21829-25-4

Trade Name	Manufacturer	Country	Year Introduced
Adalat	Bayer	W. Germany	1975
Adalat	Bayer	Italy	1976
Adalat	Bayer	Japan	1976
Adalat	Bayer	U.K.	1977
Adalate	Bayer	France	1979
Procardia	Pfizer	U.S.	1982
Alfadat	Alfa	Italy	—
Anifed	Zoja	Italy	—
Atanal	Sawai	Japan	—
Citilat	C.T.	Italy	—
Coral	Tosi	Italy	—
Corinfar	Arzneimittelwerk Dresden	E. Germany	—
Nifedacor	Schiapparelli	Italy	—
Nifedin	Gentili	Italy	—
Nifelat	Sidus	Argentina	—
Oxcord	Biosintetica	Brazil	—

Raw Materials

2-Nitrobenzaldehyde
Acetoacetic acid methyl ester
Ammonia

Manufacturing Process

45 grams 2-nitrobenzaldehyde, 80 cc acetoacetic acid methyl ester, 75 cc methanol and 32 cc ammonia are heated under reflux for several hours, filtered off, cooled and, after

suction-filtration, 75 grams of yellow crystals of MP 172° to 174°C are obtained, according to U.S. Patent 3,485,847.

References

Merck Index 6374

DFU 6 (7) 427 (1981)

Kleeman & Engel p. 633

PDR p. 1423

OCDS Vol. 2 p. 283 (1980)

DOT 8 (11) 438 (1972); 11 (4) 154 (1975) & 19 (3) 171 (1983)

I.N. p. 673

REM p. 862

Bossert, F. and Vater, W.; U.S. Patent 3,485,847; December 23, 1969; assigned to Farbenfabriken Bayer AG, Germany

Bossert, F. and Vater, W.; U.S. Patent 3,488,359; January 6, 1970; assigned to Farbenfabriken Bayer AG, Germany

Bossert, F. and Vater, W.; U.S. Patent 3,511,837; May 12, 1970; assigned to Farbenfabriken Bayer AG, Germany

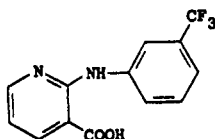
NIFLUMIC ACID

Therapeutic Function: Antiinflammatory

Chemical Name: 2-[[3-(trifluoromethyl)phenyl]amino]-3-pyridinecarboxylic acid

Common Name: 2-[3-(trifluoromethyl)anilino] nicotinic acid

Structural Formula:



Chemical Abstracts Registry No.: 4394-00-7

Trade Name	Manufacturer	Country	Year Introduced
Nifluril	U.P.S.A.	France	1968
Actol	Von Heyden	W. Germany	1971
Fiaminon	Squibb	Italy	1979
Forenol	Roemmers	Argentina	—
Landruma	Landerlan	Spain	—
Nifluran	Eczacibasi	Turkey	—
Niflux	Labofarma	Brazil	—

Raw Materials

Nicotinic acid
m-Trifluoromethylaniline
Potassium iodide

Manufacturing Process

Niflumic acid is prepared as follows: Nicotinic acid, m-trifluoromethylaniline, and potassium iodide are intimately mixed and heated on an oil bath at 140°C. The mixture melts

to give a dark red liquid. The temperature of the oil bath is allowed to fall to 100°C and is maintained at this temperature for an hour and a half. The mixture puffs up and forms a yellow crystalline mass. After cooling to ordinary temperature, this mass is ground up in a mortar and extracted several times with small volumes of ether to remove excess m-trifluoromethylaniline. The residue is then washed twice with 10 ml of distilled water to remove m-trifluoromethylaniline hydrochloride and potassium iodide, and finally twice with 10 ml of 95% alcohol to remove colored resinous contaminants. After drying at 100°C, 2-(m-trifluoromethylanilino)nicotinic acid is obtained as pale yellow needles (from 70% ethanol) melting at 204°C (Kofler block).

References

Merck Index 6377

Kleeman & Engel p. 634

OCDS Vol. 1 p. 256 (1977)

DOT 4 (2) 82 (1968)

I.N. p. 34

Hoffmann, C. and Faure, A.; U.S. Patent 3,415,834; December 10, 1968; assigned to Societe anonyme dite: Laboratoires UPSA, France

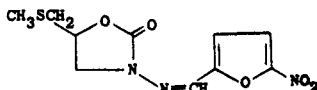
NIFURATEL

Therapeutic Function: Vaginal antiinfective

Chemical Name: 5-[(methylthio)methyl]-3-[[[(5-nitro-2-furanyl)methylene]amino]-2-oxazolidinone

Common Name: Methylmercadone

Structural Formula:



Chemical Abstracts Registry No.: 4936-47-4

Trade Name	Manufacturer	Country	Year Introduced
Macmiror	Poli	Italy	1965
Inimur	Woelm	W. Germany	1969
Omnes	Fumouze	France	1971
Magmilor	Calmic	U.K.	—
Polmiror	Poli	Italy	—
Tydantil	Poli	Italy	—

Raw Materials

Methyl mercaptan	Epichlorohydrin
Hydrazine hydrate	Diethyl carbonate
5-Nitro-2-furaldehyde	

Manufacturing Process

In an initial step of reactions, methyl mercaptan is reacted with epichlorohydrin to give 1-chloro-3-methylthio-2-propanol. That is reacted with hydrazine hydrate to give 3-methylmercapto-2-hydroxypropyl hydrazine.

11.8 grams of diethyl carbonate (0.1 mols) and a solution of sodium methoxide prepared from 0.12 gram of sodium in 4 cc of anhydrous methanol, were added to 13.2 grams of 3-methylmercapto-2-hydroxypropyl hydrazine. After the reaction vessel had been fitted with a Liebig condenser, the reaction mixture was heated by means of an oil bath which was gradually heated up to 110°C, to remove first methyl alcohol and then ethyl alcohol formed during the reaction. After about two-thirds of the theoretical amount of ethyl alcohol had been distilled off, the heating was discontinued and the reaction mixture was diluted with 50 cc of ethyl alcohol and poured into a 5-nitro-2-furfuraldehyde solution prepared by boiling for 30 minutes 0.1 mol of nitrofurfuraldehyde diacetate in 100 ml of ethyl alcohol and 50 ml of 1:10 sulfuric acid.

A yellow crystalline precipitate was immediately formed, which, after crystallization from acetic acid, melted at 182°C and consisted of N-(5-nitro-2-furfurylidene)-3-amino-5-methylmercaptomethyl-2-oxazolidinone.

References

Merck Index 6380

Kleeman & Engel p. 635

I.N. p. 674

Polichimica Sap, SpA, Italy; British Patent 969,126; September 9, 1964

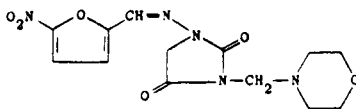
NIFURFOLINE

Therapeutic Function: Antibacterial

Chemical Name: 3-(4-Morpholinylmethyl)-1-[[(5-nitro-2-furanyl)-methylene] amino]-2,4-imidazolidinedione

Common Name: —

Structural Formula:



Chemical Abstracts Registry No.: 3363-58-4

Trade Name	Manufacturer	Country	Year Introduced
Furobactil	Carrion	France	1974
Urbac	Merck-Clevenot	France	—

Raw Materials

Nitrofurantoin
Formaldehyde
Morpholine

Manufacturing Process

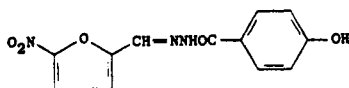
20 g of nitrofurantoin are placed in 100 cc of dimethylformamide and the solution is heated to 75°C to 80°C. This temperature is maintained and 100 cc of 40% formaldehyde are added, followed by 10 g of freshly distilled morpholine. The heating is continued for one hour, the mixture cooled and filtered and the precipitate obtained is washed with 95% alcohol. 20 g of the desired product are obtained as yellow crystals which melt at 206°C.

References

Merck Index 6381

I.N. p. 674

Laboratorios del Dr. Esteve S.A.; British Patent 1,245,095; September 2, 1971

NIFUROXAZIDE**Therapeutic Function:** Antiseptic (intestinal)**Chemical Name:** 4-Hydroxybenzoic acid [(5-nitro-2-furanyl)methylene] hydrazide**Common Name;** —**Structural Formula:****Chemical Abstracts Registry No.:** 965-52-6

Trade Name	Manufacturer	Country	Year Introduced
Ercefuryl	Carriere	France	1964
Pentofuryl	Karlspharma	W. Germany	1978
Antinal	Roques	France	—
Dicoferin	Andrade	Portugal	—
Enterokod	Genekod	France	—
Mucifural	Robert et Carriere	France	—

Raw Materials

4-Hydroxybenzhydrazide

5-Nitrofurfural

Manufacturing Process

13 g (0.1 mol) of 4-hydroxybenzhydrazide were dissolved in a boiling mixture of 100 ml of water and an equal volume of dimethylformamide. 15.5 g (0.11 mol) of 5-nitrofurfural dissolved in 31 ml of dimethylformamide were added to this hot solution, and the mixture was stirred and brought to the boiling point.

The mixture was then allowed to stand for fifteen hours. The precipitate was separated, washed twice with 100 ml of water, and recrystallized by dissolving it in 250 ml of hot pyridine and pouring this solution into 250 ml of water.

The 5-nitrofurfurylidene hydrazide of 4-hydroxybenzoic acid obtained was washed with water and methanol and was dried at a moderate temperature. It weighed 23 g (83.7% yield), and melted at 298°C. The percentage nitrogen determined by the micro-Dumas method was 15.41% (theory 15.27%).

References

Merck Index 6383

Kleeman & Engel p. 636

I.N. p. 675

Carron, M.C.E.; U.S. Patent 3,290,213; December 6, 1966; assigned to S.A. des Laboratoires Robert et Carriere (France)