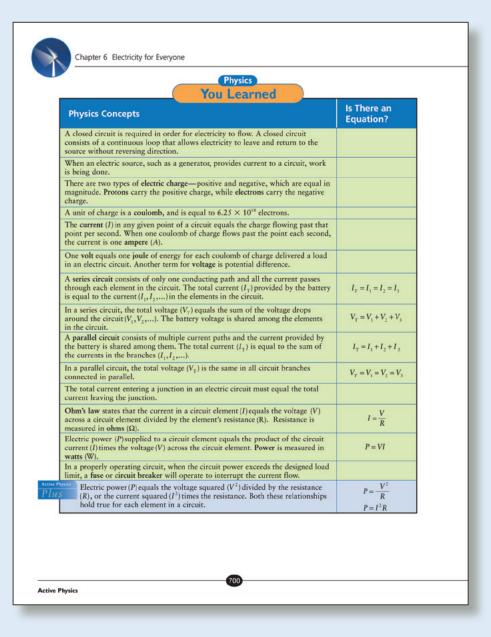
Chapter Assessment

Physics You Learned

A table listing the physics content supporting the learning outcomes of this chapter is provided to assist in reviewing and summarizing the material, and evaluating students' understanding. Consider having students review the concepts as a class, in groups, or individually.

Emphasize to students that they should also use this table as a means for reviewing for the practice exam at the end of the chapter, which will help them to further evaluate their understanding of the concepts presented. Remind them that this list can also be used as a checklist of the physics principles that should be incorporated into their *Chapter Challenge*.



An open switch will stop the flow of current in a circuit element when it is placed in series with that element.	
When measuring voltage in a circuit element, the voltmeter is always connected in parallel with that element.	
When measuring current in a circuit element, the ammeter is always placed in series with that circuit element.	
In a series circuit, the total resistance (R_{τ}) for resistors connected in series is equal to the sum of the individual resistances $(R_{\tau}, R_{2},)$. The more resistors added in series, the higher the total circuit resistance and the lower the total circuit current for a fixed voltage.	$R_{\mathrm{T}} = R_1 + R_2 + R$
In a parallel circuit, the inverse of the total resistance $\left(\frac{1}{R_{\top}}\right)$ is equal to the sum of the inverses of the individual resistances $\left(\frac{1}{R_1},\frac{1}{R_2},\ldots\right)$. The more resistors added in parallel, the lower the total circuit resistance and the higher the total circuit current for a fixed voltage.	$\frac{1}{R_{\top}} = \frac{1}{R_1} + \frac{1}{R_2} +$
The heat gained $(Q_{\rm gained})$ by one object in a closed system is equal to the heat lost $(Q_{\rm lost})$ by another object in a closed system. The law of conservation of energy states that for a closed system, no energy is lost or gained.	$Q_{gained} = Q_{lost}$
The specific heat of a material (c) is a measure of the amount of heat which must be added per gram to increase the temperature by one degree.	
The amount of heat gained or lost (ΔQ) by an object is equal to its mass (m) times its specific heat (c) times its change in temperature (ΔT) .	$\Delta Q = mc\Delta T$
Temperature is a measure of the average random kinetic energy of a material's molecules.	
The first law of thermodynamics states that heat added to a system (ΔQ) is equal to the increase in the internal energy (ΔU) of the system plus any work (W) done by the system. When heat is added, it increases the system's temperature, does work, or both.	$\Delta Q = \Delta U + W$
The second law of thermodynamics states that systems naturally move toward increasing entropy, or greater disorder. The flow of time in the universe is in the direction of increasing entropy.	
Efficiency equals the useful energy output divided by the total energy input. Efficiency is a measure of how much useful output energy is obtained from the input energy required to make a system operate.	Efficiency = useful energy out total energy input
Electric energy (E) equals electric power (P) multiplied by the operating time (t) . Electric energy is measured in kilowatt-hours (kWh) . One kilowatt hour is equal to $3,600,000$ joules (J) .	E = Pt $E = (VI)t$
Heat may be transferred through three processes – convection, conduction and radiation. Convection transfers heat to surrounding molecules by flow in fluids. Conduction transfers heat through direct contact and radiation transfers heat via electromagnetic waves. Radiation does not require a medium for transmission and therefore can occur in a vacuum.	