



















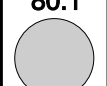



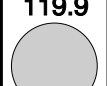




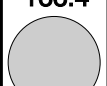





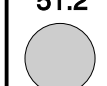











Ionization Energies of Isoelectronic Species

(Electron Volts)

				24.6
				 He
			5.4	75.6
				 Li ¹⁺
		9.3	18.2	153.9
				 Be ²⁺
	8.3	25.1	37.9	259.2
				 B ³⁺
11.3	24.4	47.8	64.4	391.9
				 C ⁴⁺

				21.6
				 Ne
			5.1	47.3
				 Na ¹⁺
		7.6	15.0	80.1
				 Mg ²⁺
	6.0	18.8	28.4	119.9
				 Al ³⁺
8.2	16.3	33.5	45.1	166.4
				 Si ⁴⁺

				15.8
				 Ar
			4.3	31.8
				 K ¹⁺
		6.1	11.9	51.2
				 Ca ²⁺
	6.5	12.8	24.8	73.9
				 Sc ³⁺
6.8	13.6	27.5	43.2	99.8
				 Ti ⁴⁺

(Columns are isoelectronic species.)