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REACTIVITIES, REAGENTS, AND REACTIVITY CHARTS

REACTIVITIES

In the selection of a protective group, it is of paramount importance to know the reactivity of the resulting protected functionality toward various reagents and reaction conditions. The number of reagents available to the organic chemist is large; approximately 8000 reagents are reviewed in the excellent series of books by the Fiesers.¹ In an effort to assess the effect of a wide variety of standard types of reagents and reaction conditions on the different possible protected functionalities, 108 prototype reagents have been selected and grouped into 16 categories:²

- A. Aqueous
- B. Nonaqueous Bases
- C. Nonaqueous Nucleophiles
- D. Organometallic
- E. Catalytic Reduction
- F. Acidic Reduction
- G. Basic or Neutral Reduction
- H. Hydride Reduction
- I. Lewis Acids
- J. Soft Acids
- K. Radical Addition

- L. Oxidizing Agents
- M. Thermal Reactions
- N. Carbenoids
- O. Miscellaneous
- P. Electrophiles

These 108 reagents are used in the Reactivity Charts that have been prepared for each class of protective groups. The reagents and some of their properties are described on the following pages.

REAGENTS

A. AQUEOUS

- | | |
|-------------------|--|
| 1. pH < 1, 100° | Refluxing HBr |
| 2. pH < 1 | 1 <i>N</i> HCl |
| 3. pH 1 | 0.1 <i>N</i> HCl |
| 4. pH 2–4 | 0.01 <i>N</i> HCl; 1–0.01 <i>N</i> HOAc |
| 5. pH 4–6 | 0.1 <i>N</i> H ₃ BO ₃ ; phosphate buffer;
HOAc–NaOAc |
| 6. pH 6–8.5 | H ₂ O |
| 7. pH 8.5–10 | 0.1 <i>N</i> HCO ₃ ⁻ ; 0.1 <i>N</i> OAc ⁻ ; satd.
CaCO ₃ |
| 8. pH 10–12 | 0.1 <i>N</i> CO ₃ ²⁻ ; 1–0.01 <i>N</i> NH ₄ OH;
0.01 <i>N</i> NaOH; satd Ca(OH) ₂ |
| 9. pH > 12 | 1–0.1 <i>N</i> NaOH |
| 10. pH > 12, 150° | |

B. NONAQUEOUS BASES

- | | |
|--|-------------------|
| 11. NaH | |
| 12. (C ₆ H ₅) ₃ CNa | $pK_a = 32$ |
| 13. [C ₁₀ H ₈] ⁻ · Na ⁺ | $pK_a \cong 37$ |
| 14. CH ₃ SOCH ₂ ⁻ Na ⁺ | $pK_a = 35$ |
| 15. KO– <i>t</i> -C ₄ H ₉ | $pK_a = 19$ |
| 16. LiN(<i>i</i> -C ₃ H ₇) ₂ | (LDA) $pK_a = 36$ |
| 17. Pyridine; Et ₃ N | $pK_a = 5; 10$ |
| 18. NaNH ₂ ; NaNHR | $pK_a = 36$ |

C. NONAQUEOUS NUCLEOPHILES

- | | |
|---|-------------|
| 19. NaOCH ₃ /CH ₃ OH, 25° | $pK_a = 16$ |
| 20. Enolate anion | $pK_a = 20$ |
| 21. NH ₃ ; RNH ₂ ; RNHOH | $pK_a = 10$ |

22. RS^- ; N_3^- ; SCN^-
 23. OAc^- ; X^- $\text{p}K_a = 4.5$
 24. NaCH , pH 12
 25. HCN , cat. CN^- , pH 6 $\text{p}K_a = 9$. For cyanohydrin formation

D. ORGANOMETALLIC

26. RLi
 27. RMgX
 28. Organozinc Reformatsky reaction. Similar:
 R_2Cu ; R_2Cd
 29. Organocopper R_2CuLi
 30. Wittig; ylide Includes sulfur ylides

E. CATALYTIC REDUCTION

31. $\text{H}_2/\text{Raney Ni}$
 32. H_2/Pt , pH 2–4
 33. $\text{H}_2/\text{Pd-C}$
 34. $\text{H}_2/\text{Lindlar}$
 35. $\text{H}_2/\text{Rh-C}$ or $\text{H}_2/\text{Rh-Al}_2\text{O}_3$ Avoids hydrogenolysis of
 benzyl ethers

F. ACIDIC REDUCTION

36. Zn/HCl
 37. Zn/HOAc ; SnCl_2/HCl
 38. Cr(II) , pH 5

G. BASIC OR NEUTRAL REDUCTION

39. Na/NH_3
 40. Al(Hg)
 41. SnCl_2/Py
 42. H_2S or HSO_3^-

H. HYDRIDE REDUCTION

43. LiAlH_4
 44. $\text{Li-}i\text{-s-Bu}_3\text{BH}$, -50° Li-Selectride
 45. $[(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)]_2\text{BH}$ Disiamylborane
 46. B_2H_6 , 0°
 47. NaBH_4
 48. $\text{Zn(BH}_4)_2$ Neutral reduction
 49. NaBH_3CN , pH 4–6
 50. $(i\text{-C}_4\text{H}_9)_2\text{AlH}$, -60° Dibal
 51. $\text{Li(O-}i\text{-C}_4\text{H}_9)_3\text{AlH}$, 0°

I. LEWIS ACIDS (ANHYDROUS CONDITIONS)

52. AlCl_3 , 80°
 53. AlCl_3 , 25°
 54. SnCl_4 , 25° ; $\text{BF}_3 \cdot \text{Et}_2\text{O}$
 55. LiClO_4 ; MgBr_2 For epoxide rearrangement
 56. TsOH , 80° Catalytic amount
 57. TsOH , 0° Catalytic amount

J. SOFT ACIDS

58. Hg(II)
 59. Ag(I)
 60. Cu(II)/Py For example, for Glaser coupling

K. RADICAL ADDITION

61. HBr/initiator "Acidic" HX addition; acidity $\cong \text{TsOH}$, 0°
 62. HX/initiator Neutral HX addition; X = P, S, Se, Si
 63. NBS/CCl_4 , $h\nu$ or heat Allylic bromination
 64. CHBr_3 ; BrCCl_3 ; $\text{CCl}_4/\text{In}\cdot$ Carbon-halogen addition

L. OXIDIZING AGENTS

65. OsO_4
 66. KMnO_4 , 0° , pH 7
 67. O_3 , -50°
 68. RCO_3H , 0° Epoxidation of olefins; prototype for $\text{H}_2\text{O}_2/\text{H}^+$
 69. RCO_3H , 50° Baeyer-Villiger oxidation of hindered ketones
 70. CrO_3/Py Collins oxidation
 71. CrO_3 , pH 1 Jones oxidation
 72. $\text{H}_2\text{O}_2/\text{OH}^-$, pH 10–12
 73. Quinone Dehydrogenation
 74. $^1\text{O}_2$ Singlet oxygen
 75. CH_3SOCH_3 , 100° (DMSO); HCO_3^- may be added to maintain neutrality
 76. NaOCl , pH 10
 77. Aq. NBS Nonradical conditions
 78. I_2
 79. $\text{C}_6\text{H}_5\text{SCl}$; $\text{C}_6\text{H}_5\text{SeX}$
 80. Cl_2 ; Br_2
 81. $\text{MnO}_2/\text{CH}_2\text{Cl}_2$
 82. NaIO_4 , pH 5–8
 83. SeO_2 , pH 2–4
 84. SeO_2/Py In EtOH/cat. Py

- | | |
|-----------------------------|--|
| 85. $K_3Fe(CN)_6$, pH 7–10 | Phenol coupling |
| 86. Pb(IV), 25° | Glycol and α -hydroxy acid cleavage |
| 87. Pb(IV), 80° | Oxidative decarboxylation |
| 88. $Tl(NO_3)_3$, pH 2 | Oxidative rearrangement of olefins |

M. THERMAL REACTIONS

- | | |
|----------|---|
| 89. 150° | Some Cope rearrangements and
Cope eliminations |
| 90. 250° | Claisen or Cope rearrangement |
| 91. 350° | Ester cracking; Conia “ene” reaction |

N. CARBENOIDS

- | | |
|--------------------------------|------------------------|
| 92. $:CCl_2$ | |
| 93. $N_2CHCO_2C_2H_5/Cu$, 80° | |
| 94. $CH_2I_2/Zn-Cu$ | Simmons–Smith addition |

O. MISCELLANEOUS

- | | |
|-------------------------------|--|
| 95. $n-Bu_3SnH$ /initiator | |
| 96. $Ni(CO)_4$ | |
| 97. CH_2N_2 | |
| 98. $SOCl_2$ | |
| 99. Ac_2O , 25° | Acetylation |
| 100. Ac_2O , 80° | Dehydration |
| 101. DCC | Dicyclohexylcarbodiimide,
$C_6H_{11}N=C=NC_6H_{11}$ |
| 102. CH_3I | |
| 103. $(CH_3)_3O^+BF_4^-$ | Or CH_3OSO_2F =Magic Methyl:
SEVERE POISON |
| 104. 1. $LiN-i-Pr_2$; 2. MeI | For C-alkylation |
| 105. 1. K_2CO_3 ; 2. MeI | For O-alkylation |

P. ELECTROPHILES

- | | |
|-----------------------|-------------------------------|
| 106. RCHO | |
| 107. $RCOCl$ | |
| 108. C^+ ion/olefin | For cation-olefin cyclization |

REACTIVITY CHARTS

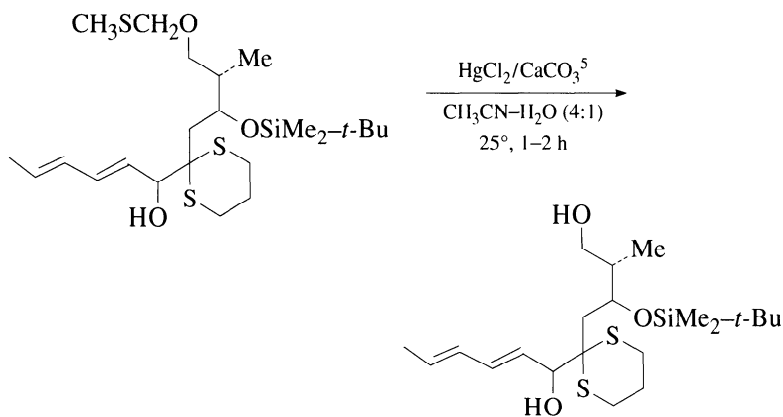
One requirement of a protective group is stability to a given reaction. The charts that follow were prepared as a guide to relative reactivities and thereby as an aid in the choice of a protective group. The reactivities in the charts were estimated

by the individual and collective efforts of a group of synthetic chemists. *It is important to realize that not all the reactivities in the charts have been determined experimentally and considerable conjecture has been exercised.* For those cases in which a literature reference was available concerning the use of a protective group and one of the 108 prototype reagents, the reactivity is printed in italic type. However, an exhaustive search for such references has not been made; therefore, the absence of italic type does not imply an experimentally unknown reactivity.

There are four levels of reactivity in the charts:

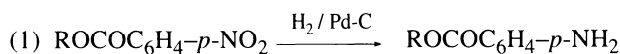
“H” (high) indicates that under the conditions of the prototype reagent, the protective group is readily removed to regenerate the original functional group.

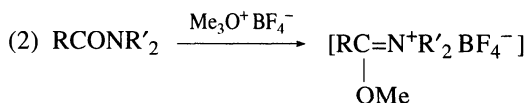
“M” (marginal) indicates that the stability of the protected functionality is marginal and depends on the exact parameters of the reaction. The protective group may be stable, may be cleaved slowly, or may be unstable to the conditions. Relative rates are always important, as illustrated in the following example⁵ (in which a monothioacetal is cleaved in the presence of a dithiane), and may have to be determined experimentally.



“L” (low) indicates that the protected functionality is stable under the reaction conditions.

“R” (reacts) indicates that the protected compound reacts readily, but that the original functional group is not restored. The protective group may be changed to a new protective group (eq. 1) or to a reactive intermediate (eq. 2), or the protective group may be unstable to the reaction conditions and react further (eq. 3).





The reactivities in the charts refer *only* to the protected functionality, not to atoms adjacent to the functional group; for example, RCOOEt $\xrightarrow{\text{LDA}}$: “L” (low) reactivity of PG(Et). However, if the protected functionality is R₂CHCOOEt, this substrate obviously *will* react with LDA. Reactivity of the entire substrate must be evaluated by the chemist.

Five reagents [#25: HCN, pH 6; #88: Tl(NO₃)₃; #103: Me₃O⁺ BF₄⁻; #104: LiN-*i*-Pr₂/MeI; and #105: K₂CO₃/MeI] were added after some of the charts had been completed; reactivities to these reagents are not included for all charts.

The number used to designate a protective group (PG) in a Reactivity Chart is the same as that used in the body of the text in the *first* edition.

Protective group numbers in the Reactivity Charts are not continuous, since not all of the protective groups described in the text are included in the charts. The protective groups that are included in the Reactivity Charts are, in general, those that have been used most widely; consequently, considerable experimental information is available for them.

The Reactivity Charts were prepared in collaboration with the following chemists, to whom we are most grateful: John O. Albright, Dale L. Boger, Dr. Daniel J. Brunelle, Dr. David A. Clark, Dr. Jagabandhu Das, Herbert Estreicher, Anthony L. Feliu, Dr. Frank W. Hobbs, Jr., Paul B. Hopkins, Dr. Spencer Knapp, Dr. Pierre Lavallée, John Munroe, Jay W. Ponder, Marcus A. Tius, Dr. David R. Williams, and Robert E. Wolf, Jr.

¹ L. F. Fieser and M. Fieser, *Reagents for Organic Synthesis*, Wiley-Interscience, New York, 1967, Vol. 1; M. Fieser and L. F. Fieser, Vols. 2–7, 1969–1979; M. Fieser, Vols. 8–17, 1980–1994.

² The categories and prototype reagents used in this study are an expansion of an earlier set of 11 categories and 60 prototype reagents,³ originally compiled for use in LHASA⁴ (Logic and Heuristics Applied to Synthetic Analysis), a long-term research program at Harvard University for Computer-Assisted Synthetic Analysis.

³ E. J. Corey, H. W. Orf, and D. A. Pensak, *J. Am. Chem. Soc.*, **98**, 210 (1976).

⁴ Selected references include E. J. Corey, *Quart. Rev., Chem. Soc.*, **25**, 455 (1971); H. W. Orf, Ph. D. Thesis, Harvard University, 1976.

⁵ E. J. Corey and M.G. Bock, *Tetrahedron Lett.*, 2643 (1975).

Reactivity Chart 1. Protection for Hydroxyl Group: Ethers

1. Methyl Ether
2. Methoxymethyl Ether (MOM)
3. Methylthiomethyl Ether (MTM)
6. 2-Methoxyethoxymethyl Ether (MEM)
8. Bis(2-chloroethoxy)methyl Ether
9. Tetrahydropyranyl Ether (THP)
11. Tetrahydrothiopyranyl Ether
12. 4-Methoxytetrahydropyranyl Ether
13. 4-Methoxytetrahydrothiopyranyl Ether
15. Tetrahydrofuranlyl Ether
16. Tetrahydrothiofuranlyl Ether
17. 1-Ethoxyethyl Ether
18. 1-Methyl-1-methoxyethyl Ether
21. 2-(Phenylselenyl)ethyl Ether
22. *t*-Butyl Ether
23. Allyl Ether
26. Benzyl Ether
28. *o*-Nitrobenzyl Ether
35. Triphenylmethyl Ether
36. α -Naphthylidiphenylmethyl Ether
37. *p*-Methoxyphenyldiphenylmethyl Ether
41. 9-(9-Phenyl-10-oxo)anthryl Ether (Tritylone)
43. Trimethylsilyl Ether (TMS)
45. Isopropyldimethylsilyl Ether
46. *t*-Butyldimethylsilyl Ether (TBDMS)
48. *t*-Butyldiphenylsilyl Ether
51. Tribenzylsilyl Ether
53. Triisopropylsilyl Ether

(See chart, pp. 709–711.)

Reactivity Chart 1. Protection for the Hydroxyl Group: Ethers (Continued)

PG	78	79	80	81	82	83	84	85	86	87	88	M.				N.				O. MISCELLANEOUS				106	107	108						
	I_2	PhSeX; PhSCl	Br_2/Cl_2	NO_2/CH_2Cl_2	$NaIO_4$ pH 5-8	SeO_2 pH 2-4	SeO_2/Py	$K_2Fe(CN)_6$, pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	CCl_2	M_2CHCO_2R/Cu	$CH_2I_2/Zn(Cu)$	R_3SnH/In	Nf(CO) ⁴	CH_2N_2	$SOCl_2$	Ac_2O , 25°	Ac_2O , 80°	DCC	MeI	$Me_3O^+BF_4^-$	1. LDA 2. MeI	1. K_2CO_3 2. MeI	RCHO	RCOCl	$C^+/olefin$	
1	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
2	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
3	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
6	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
8	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
9	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
11	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
12	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
13	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
15	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
16	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
17	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
18	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
21	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
22	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
23	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
26	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
28	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
35	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
36	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
37	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
41	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
43	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
45	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
46	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
48	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
51	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
53	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L

Reactivity Chart 2. Protection for Hydroxyl Group: Esters

1. Formate Ester
3. Acetate Ester
6. Trichloroacetate Ester
10. Phenoxyacetate Ester
19. Isobutyrate Ester
22. Pivaloate Ester
23. Adamantoate Ester
27. Benzoate Ester
31. 2,4,6-Trimethylbenzoate (Mesitoate) Ester
34. Methyl Carbonate
36. 2,2,2-Trichloroethyl Carbonate
39. Allyl Carbonate
41. *p*-Nitrophenyl Carbonate
42. Benzyl Carbonate
46. *p*-Nitrobenzyl Carbonate
47. *S*-Benzyl Thiocarbonate
48. *N*-Phenylcarbamate
51. Nitrate Ester
53. 2,4-Dinitrophenylsulfenate Ester

(See chart, pp. 713–715.)

Reactivity Chart 2. Protection for the Hydroxyl Group: Esters

PG	pH < 1, 100°			A. AQUEOUS						B. BASIC					C. NUCLEOPHILIC				Organozinc			Organocopper		Wittig; ylide		H ₂ /Raney (Ni)				H ₂ /Pt pH 2-4		H ₂ /Pd		H ₂ /Lindlar		H ₂ /Rh		Zn/HOAc		Zn/HCl		Cr(II), pH 5																						
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38																										
1	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L																					
3	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L																		
6	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L															
10	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L													
19	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L													
22	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	M	M	M	M	M	M	M	M	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L															
23	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L												
27	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L										
31	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L									
34	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L								
36	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L							
39	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L							
41	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L						
42	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L					
46	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L					
47	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L					
48	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L				
51	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L		
53	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	L	L	L	L	L	L	L	L	L	L	L	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	L	L	L	L

Reactivity Chart 2. Protection for the Hydroxyl Group: Esters (Continued)

PG	G.	H. HYDRIDE REDN.	I.								K.				L. OXIDANTS																																												
	Na/NH ₃	Al(Hg)	SnCl ₂ /Py	H ₂ O ₂ /H ₂ S	LAH ^a	Li-S-Bu ³ BH	(C ³ H ₁₁) ₂ BH	B ² H ₆ , 0°	NABH ₄	Zn(BH ₄) ₂	NABH ₃ CN pH 4-6	t-Bu ₂ AlH	Li(O ⁱ Bu) ₃ AlH	AlCl ₃ , 80°	AlCl ₃ , 25°	SnCl ₄ ; BF ₃	LiClO ₄ ; MgBr ₂	TsOH, 80°	TsOH, 0°	Hg(II)	Ag(I)	Cu(II)/Py	HBr/In.	HX/In.	NBS/CCl ₄	Br ₂ CCl ₂ /In.	OsO ₄	KMnO ₄ , pH 7.0°	O ₃ , -50°	RCO ₂ H, 0°	RCO ₂ H, 50°	CrO ₃ /Py	CrO ₃ , pH 1	H ₂ O ₂ pH 10-12	Quinone	O ₂	DMSO, 100°	NaOCl pH 10	aq NBS																				
39	H	H	L	L	H	L	L	M	L	L	L	M	M	52	H	L	L	L	L	58	L	L	L	L	L	L	65	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L										
40	H	H	L	L	H	L	L	L	L	L	L	M	M	53	H	L	L	L	L	59	L	L	L	L	L	L	66	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L				
41	H	H	L	L	H	L	L	L	L	L	L	M	M	54	H	L	L	L	L	60	L	L	L	L	L	L	67	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
42	H	H	L	L	H	L	L	L	L	L	L	M	M	55	L	L	L	L	L	61	L	L	L	L	L	L	68	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
43	H	H	L	L	H	L	L	L	L	L	L	M	M	56	H	L	L	L	L	62	L	L	L	L	L	L	69	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
44	H	H	L	L	H	L	L	L	L	L	L	M	M	57	H	L	L	L	L	63	L	L	L	L	L	L	70	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
45	H	H	L	L	H	L	L	L	L	L	L	M	M	58	H	L	L	L	L	64	L	L	L	L	L	L	71	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
46	H	H	L	L	H	L	L	L	L	L	L	M	M	59	H	L	L	L	L	65	L	L	L	L	L	L	72	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
47	H	H	L	L	H	L	L	L	L	L	L	M	M	60	H	L	L	L	L	66	L	L	L	L	L	L	73	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
48	H	H	L	L	H	L	L	L	L	L	L	M	M	61	H	L	L	L	L	67	L	L	L	L	L	L	74	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
49	H	H	L	L	H	L	L	L	L	L	L	M	M	62	H	L	L	L	L	68	L	L	L	L	L	L	75	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
50	H	H	L	L	H	L	L	L	L	L	L	M	M	63	H	L	L	L	L	69	L	L	L	L	L	L	76	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
51	H	H	L	L	H	L	L	L	L	L	L	M	M	64	H	L	L	L	L	70	L	L	L	L	L	L	77	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
52	H	H	L	L	H	L	L	L	L	L	L	M	M	65	H	L	L	L	L	71	L	L	L	L	L	L	78	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
53	H	H	L	L	H	L	L	L	L	L	L	M	M	66	H	L	L	L	L	72	L	L	L	L	L	L	79	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	

Reactivity Chart 3. Protection for 1,2- and 1,3-Diols

1. Methylendioxy Derivative
2. Ethylidene Acetal
6. Acetonide Derivative
11. Benzylidene Acetal
13. *p*-Methoxybenzylidene Acetal
18. Methoxymethylene Acetal
20. Dimethoxymethylendioxy Derivative
28. Cyclic Carbonates
29. Cyclic Boronates

(See chart, pp. 717–719.)

Reactivity Chart 3. Protection for 1,2- and 1,3-Diols

PG	A. AQUEOUS												B. BASIC												C. NUCLEOPHILIC												D. ORGANOMET.												E. CAT. REDN.												F.																																																																																																																																																																																																																																																																																			
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50																																																																																																																																																																																																																																																																																														
	PH < 1, 100°												Ph ₃ CNa												(C ₁₀ H ₈) ⁻ Na ⁺												MeSOCH ₂ ⁻ Na ⁺												KO ⁻ t-Bu												LiN ⁻ i-Pr ₂												Py; R ₃ N												NaNH ₂												NaOMe												Enolate												NH ₃ ; RNH ₂												RS ⁻ ; N ₃ ⁻ ; SCN ⁻												OAc ⁻ ; X ⁻												NaCN; pH 12												HCN; pH 6												RTI												RMgX												Organozinc												Organo-copper												Wittig; Ylide												H ₂ /Raney (Ni)												H ₂ /Pt pH 2-4												H ₂ /Pd												H ₂ /Lindlar												H ₂ /Rh												Zn/HCl												Zn/HOAc												Cr(II), pH 5											

Reactivity Chart 3. Protection for 1,2- and 1,3-Diols (Continued)

PG	78	79	80	81	82	83	84	85	86	87	88	O. MISCELLANEOUS												P.							
1	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108
2	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
6	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
11	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
12	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
18	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
20	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
28	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
29	I ₂	phSeX; PhSCL	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Et ₂	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin

Reactivity Chart 4. Protection for Phenols and Catechols

Phenols

1. Methyl Ether
2. Methoxymethyl Ether
3. 2-Methoxyethoxymethyl Ether
4. Methylthiomethyl Ether
6. Phenacyl Ether
7. Allyl Ether
8. Cyclohexyl Ether
9. *t*-Butyl Ether
10. Benzyl Ether
11. *o*-Nitrobenzyl Ether
12. 9-Anthrylmethyl Ether
13. 4-Picolyl Ether
15. *t*-Butyldimethylsilyl Ether
16. Aryl Acetate
17. Aryl Pivaloate
18. Aryl Benzoate
19. Aryl 9-Fluorencarboxylate
20. Aryl Methyl Carbonate
21. Aryl 2,2,2-Trichloroethyl Carbonate
22. Aryl Vinyl Carbonate
23. Aryl Benzyl Carbonate
25. Aryl Methanesulfonate

Catechols

27. Methylenedioxy Derivative
28. Acetonide Derivative
30. Diphenylmethylenedioxy Derivative
31. Cyclic Borates
32. Cyclic Carbonates

(See chart, pp. 721–723.)

Reactivity Chart 4. Protection for Phenols and Catechols (Continued)

PG	I ₂	PhSeCl; PhSeCl	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ /Py	SeO ₂ pH 2-4	K ₂ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	PI(NO ₃) ₃	M.	N.	R ₂ CHCO ₂ R/Cu	R ₃ SnH/In.	NH(CO) ₂	CH ₂ N ₂	SOCI ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. IMA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin								
	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108					
	L. OXIDANTS												O. MISCELLANEOUS																							
1	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L					
2	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L				
3	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L				
4	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L			
6	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L			
7	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L			
8	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L			
9	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
10	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
11	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
12	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
13	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
15	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
16	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
17	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
18	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
19	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
20	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
21	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
22	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
23	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
25	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
27	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
28	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
30	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
31	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
32	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L

Reactivity Chart 5. Protection for the Carbonyl Group

1. Dimethyl Acetals and Ketals
3. Bis(2,2,2-trichloroethyl) Acetals and Ketals
5. 1,3-Dioxanes
6. 5-Methylene-1,3-dioxanes
7. 5,5-Dibromo-1,3-dioxanes
8. 1,3-Dioxolanes
9. 4-Bromomethyl-1,3-dioxolanes
10. 4-*o*-Nitrophenyl-1,3-dioxolanes
11. *S,S'*-Dimethyl Acetals and Ketals
19. 1,3-Dithianes
20. 1,3-Dithiolanes
24. 1,3-Oxathiolanes
26. *O*-Trimethylsilyl Cyanohydrins
29. *N,N*-Dimethylhydrazones
30. 2,4-Dinitrophenylhydrazones
33. *O*-Phenylthiomethyl Oximes
34. Substituted Methylene Derivatives
43. Bismethylenedioxy Derivatives

(See chart, pp. 725–727.)

reactivity Chart 5. Protection for the Carbonyl Group (Continued)

PG	I ₂	Phsex; PhSCI	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Py	K ₃ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	M.	N.	O. MISCELLANEOUS	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	RCOCl	C ⁺ /olefin
78	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
79	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
80	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
81	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
82	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
83	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M
84	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
85	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
86	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
87	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
88	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
89	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
90	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H
91	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
92	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
93	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
94	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
95	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R	R
96	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M
97	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
98	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
99	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
100	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
101	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
102	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
103	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
104	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M
105	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
106	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
107	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
108	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M	M
19	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
20	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H	H
24	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
26	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
29	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
30	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
33	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
34	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
43	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L

Reactivity Chart 6. Protection for the Carboxyl Group

Esters

1. Methyl Ester
2. Methoxymethyl Ester
3. Methylthiomethyl Ester
4. Tetrahydropyranyl Ester
7. Benzyloxymethyl Ester
8. Phenacyl Ester
13. *N*-Phthalimidomethyl Ester
15. 2,2,2-Trichloroethyl Ester
16. 2-Haloethyl Ester
21. 2-(*p*-Toluenesulfonyl)ethyl Ester
23. *t*-Butyl Ester
27. Cinnamyl Ester
30. Benzyl Ester
31. Triphenylmethyl Ester
33. Bis(*o*-nitrophenyl)methyl Ester
34. 9-Anthrylmethyl Ester
35. 2-(9,10-Dioxo)anthrylmethyl Ester
42. Piperonyl Ester
45. Trimethylsilyl Ester
47. *t*-Butyldimethylsilyl Ester
50. *S-t*-Butyl Ester
59. 2-Alkyl-1,3-oxazolines

Amides and Hydrazides

64. *N,N*-Dimethylamide
68. *N*-7-Nitroindoylamide
71. Hydrazides
72. *N*-Phenylhydrazide
73. *N,N'*-Diisopropylhydrazide

(See chart, pp. 729–731.)

Reactivity Chart 7. Protection for the Thiol Group

1. *S*-Benzyl Thioether
3. *S-p*-Methoxybenzyl Thioether
5. *S-p*-Nitrobenzyl Thioether
6. *S*-4-Picolyl Thioether
7. *S*-2-Picolyl *N*-Oxide Thioether
8. *S*-9-Anthrylmethyl Thioether
9. *S*-Diphenylmethyl Thioether
10. *S*-Di(*p*-methoxyphenyl)methyl Thioether
12. *S*-Triphenylmethyl Thioether
15. *S*-2,4-Dinitrophenyl Thioether
16. *S-t*-Butyl Thioether
19. *S*-Isobutoxymethyl Monothioacetal
20. *S*-2-Tetrahydropyranyl Monothioacetal
23. *S*-Acetamidomethyl Aminothioacetal
25. *S*-Cyanomethyl Thioether
26. *S*-2-Nitro-1-phenylethyl Thioether
27. *S*-2,2-Bis(carboethoxy)ethyl Thioether
30. *S*-Benzoyl Derivative
36. *S*-(*N*-Ethylcarbamate)
38. *S*-Ethyl Disulfide

(See chart, pp. 733–735.)

Reactivity Chart 7. Protection for the Thiol Group

PG	A. AQUEOUS										B. BASIC										C. NUCLEOPHILIC										D. ORGANOMET.					E. CAT. REDN.					F.				
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38							
	pH<1, 100°	pH<1	pH 1	pH 2-4	pH 4-6	pH 6-8.5	pH 8.5-10	pH 10-12	pH>12	pH>12, 150°	NaH	Ph ₃ CNa	(C ¹⁰ H ₈) ⁻ Na ⁺	MesOCH ₂ ⁻ Na ⁺	KO ⁻ t-Bu	t-Bu ⁻ Pr ₂	Py: R ₃ N	NaNH ₂	NaOMe	Enolate	NH ₃ : RNH ₂	RS ⁻ : N ₃ ⁻ : SCN ⁻	OAc ⁻ : X ⁻	NaCN, pH 12	HCN, pH 6	RTI	RMgX	Organozinc	Organocopper	Wittig: ylide	H ₂ /Raney (Ni)	H ₂ /Pt pH 2-4	H ₂ /Pd	H ₂ /Lindlar	H ₂ /Rh	Zn/HCl	Zn/HOAc	Cr(II), pH 5							
1	H	H	H	H	H	H	H	H	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
3	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
5	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
6	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
7	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
8	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
9	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
10	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
12	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
15	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
16	H	M	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
19	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
20	H	H	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
23	H	M	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
25	H	R	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
26	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
27	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
30	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
36	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		
38	H	L	L	L	L	L	L	L	M	M	L	R	R	R	R	R	R	L	L	L	L	L	L	L	L	R	R	R	R	R	R	R	R	R	R	L	L	L	L	L	L	L	L		

Reactivity Chart 7. Protection for the Thiol Group (Continued)

PG	78	79	80	81	82	83	84	85	86	87	88	M.								O. MISCELLANEOUS								106		107		108			
	1 ₂	Phsex; PhSCl	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Py	K ₂ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	150°	250°	350°	:CCl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SnH/In.	NI(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	RGHO	RCOCl	C ⁺ /olefin				
1	L	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	
3	M	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	
5	M	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
6	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
7	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
8	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
9	M	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
10	M	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
12	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
15	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
16	L	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
19	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
20	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
23	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
25	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
26	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
27	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
30	L	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
36	L	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L
38	R	L	L	L	L	L	L	L	L	L	L	L	L	L	M	R	L	R	L	L	L	L	L	L	L	L	R	R	L	L	L	L	L	L	L

Reactivity Chart 8. Protection for the Amino Group: Carbamates

1. Methyl Carbamate
5. 9-Fluorenylmethyl Carbamate
8. 2,2,2-Trichloroethyl Carbamate
11. 2-Trimethylsilylethyl Carbamate
16. 1,1-Dimethylpropynyl Carbamate
20. 1-Methyl-1-phenylethyl Carbamate
22. 1-Methyl-1-(4-biphenyl)ethyl Carbamate
24. 1,1-Dimethyl-2-haloethyl Carbamate
26. 1,1-Dimethyl-2-cyanoethyl Carbamate
28. *t*-Butyl Carbamate
30. Cyclobutyl Carbamate
31. 1-Methylcyclobutyl Carbamate
35. 1-Adamantyl Carbamate
37. Vinyl Carbamate
38. Allyl Carbamate
39. Cinnamyl Carbamate
44. 8-Quinolyl Carbamate
45. *N*-Hydroxypiperidinyl Carbamate
47. 4,5-Diphenyl-3-oxazolin-2-one
48. Benzyl Carbamate
53. *p*-Nitrobenzyl Carbamate
55. 3,4-Dimethoxy-6-nitrobenzyl Carbamate
58. 2,4-Dichlorobenzyl Carbamate
65. 5-Benzisoxazolymethyl Carbamate
66. 9-Anthrylmethyl Carbamate
67. Diphenylmethyl Carbamate
71. Isonicotinyl Carbamate
72. *S*-Benzyl Carbamate
75. *N*-(*N*'-Phenylaminothiocarbonyl) Derivative

(See chart, pp. 737–739.)

Reactivity Chart 8. Protection for the Amino Group: Carbamates (Continued)

PG	78	79	80	81	82	83	84	85	86	87	88	89	90	91	M.	92	93	94	N.	95	96	97	98	99	100	101	102	103	104	105	106	107	108	
	I ₂	PhSO ₂ ; PhSO ₂ Cl	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NaIO ₄ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Py	K ₂ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Pb(NO ₃) ₂	150°	250°	350°		ICl ₂	N ₂ CHCO ₂ R/Cu	CH ₂ I ₂ /Zn(Cu)	R ₃ SiH/In.	Nf(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BPh ₄ ⁻	1. LDA 2. MeI	1. K ₂ CO ₃ 2. MeI	KCHO	ROCCl	C ⁺ /olefin		
1	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L		
5	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
8	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
11	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
16	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
20	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
22	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
24	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
26	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
28	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
30	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
31	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
35	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
37	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
38	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
39	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
44	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
45	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
47	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
48	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
53	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
55	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
58	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
65	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
66	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
67	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
71	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
72	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
75	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L

**Reactivity Chart 9. Protection for the Amino Group:
Amides**

1. *N*-Formyl
2. *N*-Acetyl
3. *N*-Chloroacetyl
5. *N*-Trichloroacetyl
6. *N*-Trifluoroacetyl
7. *N*-*o*-Nitrophenylacetyl
8. *N*-*o*-Nitrophenoxyacetyl
9. *N*-Acetoacetyl
12. *N*-3-Phenylpropionyl
13. *N*-3-(*p*-Hydroxyphenyl)propionyl
15. *N*-2-Methyl-2-(*o*-nitrophenoxy)propionyl
16. *N*-2-Methyl-2-(*o*-phenylazophenoxy)propionyl
17. *N*-4-Chlorobutyryl
19. *N*-*o*-Nitrocinnamoyl
20. *N*-Picolinoyl
21. *N*-(*N*'-Acetylmethionyl)
23. *N*-Benzoyl
29. *N*-Phthaloyl
31. *N*-Dithiasuccinoyl

(See chart, pp. 741–743.)

**Reactivity Chart 10. Protection for the Amino Group:
Special –NH Protective Groups**

1. *N*-Allyl
2. *N*-Phenacyl
3. *N*-3-Acetoxypropyl
5. Quaternary Ammonium Salts
6. *N*-Methoxymethyl
8. *N*-Benzyloxymethyl
9. *N*-Pivaloyloxymethyl
12. *N*-Tetrahydropyranyl
13. *N*-2,4-Dinitrophenyl
14. *N*-Benzyl
16. *N*-*o*-Nitrobenzyl
17. *N*-Di(*p*-methoxyphenyl)methyl
18. *N*-Triphenylmethyl
19. *N*-(*p*-Methoxyphenyl)diphenylmethyl
20. *N*-Diphenyl-4-pyridylmethyl
21. *N*-2-Picolyl *N*'-Oxide
24. *N,N*'-Isopropylidene
25. *N*-Benzylidene
27. *N*-*p*-Nitrobenzylidene
28. *N*-Salicylidene
33. *N*-(5,5-Dimethyl-3-oxo-1-cyclohexenyl)
37. *N*-Nitro
39. *N*-Oxide
40. *N*-Diphenylphosphinyl
41. *N*-Dimethylthiophosphinyl
47. *N*-Benzenesulfonyl
48. *N*-*o*-Nitrobenzenesulfonyl
55. *N*-2,4,6-Trimethylbenzenesulfonyl
56. *N*-Toluenesulfonyl
57. *N*-Benzylsulfonyl
59. *N*-Trifluoromethylsulfonyl
60. *N*-Phenacylsulfonyl

(See chart, pp. 745–747.)

Reactivity Chart 10. Protection for the Amino Group: Special – NH Protective Groups
(Continued)

PG	I ₂	PhSx; PhSCl	Br ₂ ; Cl ₂	MnO ₂ /CH ₂ Cl ₂	NiO ₂ pH 5-8	SeO ₂ pH 2-4	SeO ₂ /Py	K ₂ Fe(CN) ₆ , pH 8	Pb(IV), 25°	Pb(IV), 80°	Tl(NO ₃) ₃	M.	M.	Cl ₂	N.	R ₃ SiH/In.	Ni(CO) ₄	CH ₂ N ₂	SOCl ₂	Ac ₂ O, 25°	Ac ₂ O, 80°	DCC	MeI	Me ₃ O ⁺ BF ₄ ⁻	1. EtN 2. MeI	1. K ₂ CO ₃ 2. MeI	RCHO	ROCOCl	C ⁺ /olefin			
1	L	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R		
2	L	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R		
3	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R		
5	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	
6	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
8	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
9	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
12	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
13	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
14	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
16	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
17	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
18	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
19	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
20	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
21	L	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
24	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
25	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
27	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
28	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
33	L	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
37	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	
39	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
40	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
41	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
47	R	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R
48	R	L	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R
55	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
56	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
57	L	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R	R	R	R	R	R	R	R	R	R	L	L	R	R	R	R
59	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
60	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L